

# Epipelagic nitrous oxide production offsets carbon sequestration by the biological pump

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The removal of carbon dioxide from the atmosphere by the marine biological pump is a key regulator of Earth's climate; however, the ocean also serves as a large source of nitrous oxide, a potent greenhouse gas and ozone-depleting substance. Although biological carbon sequestration and nitrous oxide production have been individually studied in the ocean, their combined impacts on net greenhouse forcing remain uncertain. Here we show that the magnitude of nitrous oxide production in the epipelagic zone of the subtropical ocean covaries with remineralization processes and thus acts antagonistically to weaken the radiative benefit of carbon removal by the marine biological pump. Carbon and nitrogen isotope tracer incubation experiments and nitrogen isotope natural abundance data indicate enhanced biological activity promotes nitrogen recycling, leading to substantial nitrous oxide production via both oxidative and reductive pathways. These shallow-water nitrous oxide sources account for nearly half of the air–sea flux and counteract 6–27% (median 9%) of the greenhouse warming mitigation achieved by carbon export via the biological pump.

The ocean plays a crucial role in the global climate system through modulating atmospheric greenhouse gases by absorbing nearly 30% of anthropogenic carbon dioxide (CO<sub>2</sub>) and releasing 20% of total nitrous oxide (N<sub>2</sub>O) emissions to the atmosphere<sup>1,2</sup>. The marine biological pump (defined as the biologically driven processes that transfer carbon from the surface ocean to the ocean's interior) is the dominant mechanism driving long-term CO<sub>2</sub> exchange across the air–sea interface and plays a critical role in regulating atmospheric CO<sub>2</sub> and climate<sup>3–5</sup>. The efficiency of the marine biological pump, defined by the ratio of carbon export to net primary production at a specific reference depth (for example, the base of the euphotic zone), is often estimated to be -10% in the global ocean<sup>5</sup>, and even lower in the subtropical oceans<sup>6,7</sup>. In the subtropical oceans, a large fraction of newly produced organic material undergoes remineralization in the upper 200 m (the epipelagic zone), resulting in rapid carbon and nitrogen transformations between organic and inorganic forms. Numerous studies on the ocean's biological pump

have focused on the magnitude and controls of carbon removal; however, the potential counter effect of N<sub>2</sub>O emissions in offsetting the radiative effect of CO<sub>2</sub> removal by the biological pump has been largely overlooked.

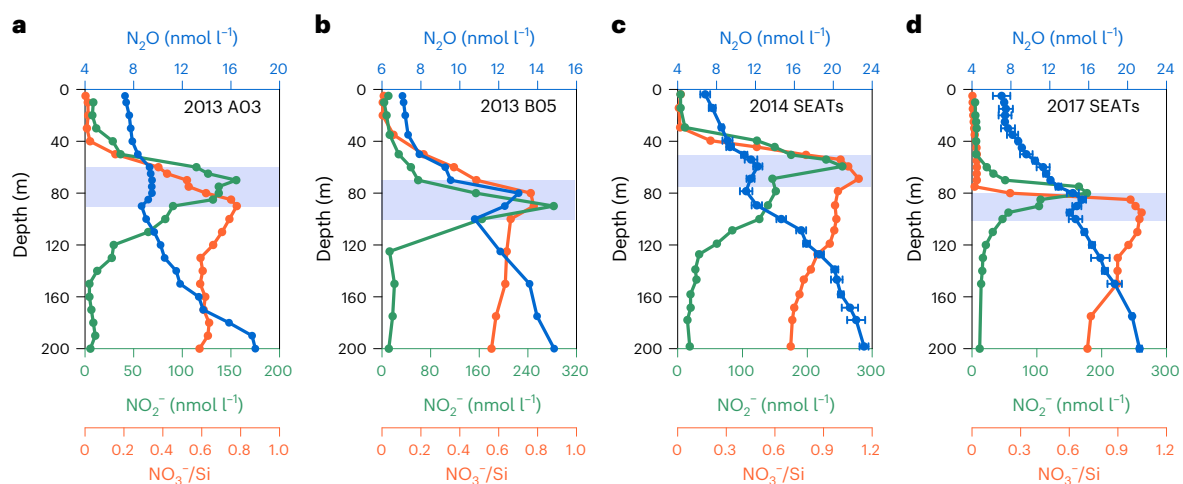
In the marine nitrogen cycle, N<sub>2</sub>O is mainly produced as a by-product of nitrification and as an intermediate during denitrification, both of which are largely controlled by organic matter supply and remineralization<sup>8</sup>. In the oxygenated ocean, nitrification is considered the dominant source of N<sub>2</sub>O (refs. <sup>8–10</sup>). However, recent studies on marine N<sub>2</sub>O suggest the sources are more complex, including both aerobic nitrification and anaerobic nitrate (NO<sub>3</sub><sup>-</sup>) and nitrite (NO<sub>2</sub><sup>-</sup>) reduction in anaerobic micro-niches associated with marine aggregates<sup>11</sup> or zooplankton guts<sup>12</sup>. Ammonia-oxidizing archaea, the dominant ammonia oxidizer in the open oceans<sup>13</sup>, utilize a hybrid N<sub>2</sub>O production pathway that is distinct from that of their bacterial counterparts, in which intermediates sourced from ammonium (NH<sub>4</sub><sup>+</sup>) and NO<sub>2</sub><sup>-</sup> co-contribute to N<sub>2</sub>O formation<sup>14</sup>. The physiological and

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**Table 1 | Primary production, export production and N<sub>2</sub>O production rate in the SCS and the NPSG**

Site	Area	PP <sup>a</sup>	EP <sup>b</sup>	e ratio (%)	N <sub>2</sub> O production <sup>c</sup>	GWP <sub>100</sub> <sup>d</sup>	Offset (%)
K1	NPSG	11.6±0.8	0.9±0.2	7.8±1.9	0.27±0.01	82.2±2.9	9.1±2.1
X1	NWP	41.7±3.9	2.0±0.4	4.8±1.1	0.57±0.03	169.8±9.2	8.5±1.7
A2	SCS	44.1±3.0	0.3±0.1	1.0±0.3	0.22±0.02	83.1±4.9	27.2±6.7
SEATs	SCS	31.9±3.3	1.2±0.1	2.7±0.3	0.28±0.02	65.0±4.0	5.6±0.6
A32	NPSG	ND	0.5±0.1	ND	0.10±0.02	31.3±3.3	6.2±1.0
Z2	NPSG	10.4±0.4	0.3±0.0	3.2±0.4	0.21±0.01	63.0±3.8	18.9±2.4
P5	ECS shelf	ND	ND	ND	0.48±0.05	138.8±13.7	ND
C5	ECS shelf	ND	ND	ND	0.38±0.02	110.0±5.0	ND

<sup>a</sup>Depth (0–125 m) integrated primary production rate (mmol C m<sup>-2</sup> d<sup>-1</sup>); see Extended Data Fig. 4. <sup>b</sup>Export production at the base of epipelagic zone (200 m) (mmol C m<sup>-2</sup> d<sup>-1</sup>). <sup>c</sup>Depth (0–200 m) integrated N<sub>2</sub>O production rate (μmol N<sub>2</sub>O m<sup>-2</sup> d<sup>-1</sup>); see Fig. 4. <sup>d</sup>The GWP<sub>100</sub> was calculated by using a global warming potential at the 100-year time horizon (μmol C m<sup>-2</sup> d<sup>-1</sup>) (GWP<sub>100</sub>, 1 mol N<sub>2</sub>O is equivalent to 300 mol CO<sub>2</sub> in retaining radiative energy)<sup>1</sup>. ND, no data; NWP, northwest Pacific; ECS shelf, East China Sea shelf.



**Fig. 1 | Depth profiles of N<sub>2</sub>O concentration, NO<sub>2</sub><sup>-</sup> and NO<sub>3</sub><sup>-</sup>/Si at stations with distinctive shallow N<sub>2</sub>O accumulation in the epipelagic zone. a–d, N<sub>2</sub>O concentration (blue dots), NO<sub>2</sub><sup>-</sup> concentration (green dots) and NO<sub>3</sub><sup>-</sup>/Si (orange dots) at stations A03 (2013; a), B05 (2013; b), the Southeast Asian Time-series**

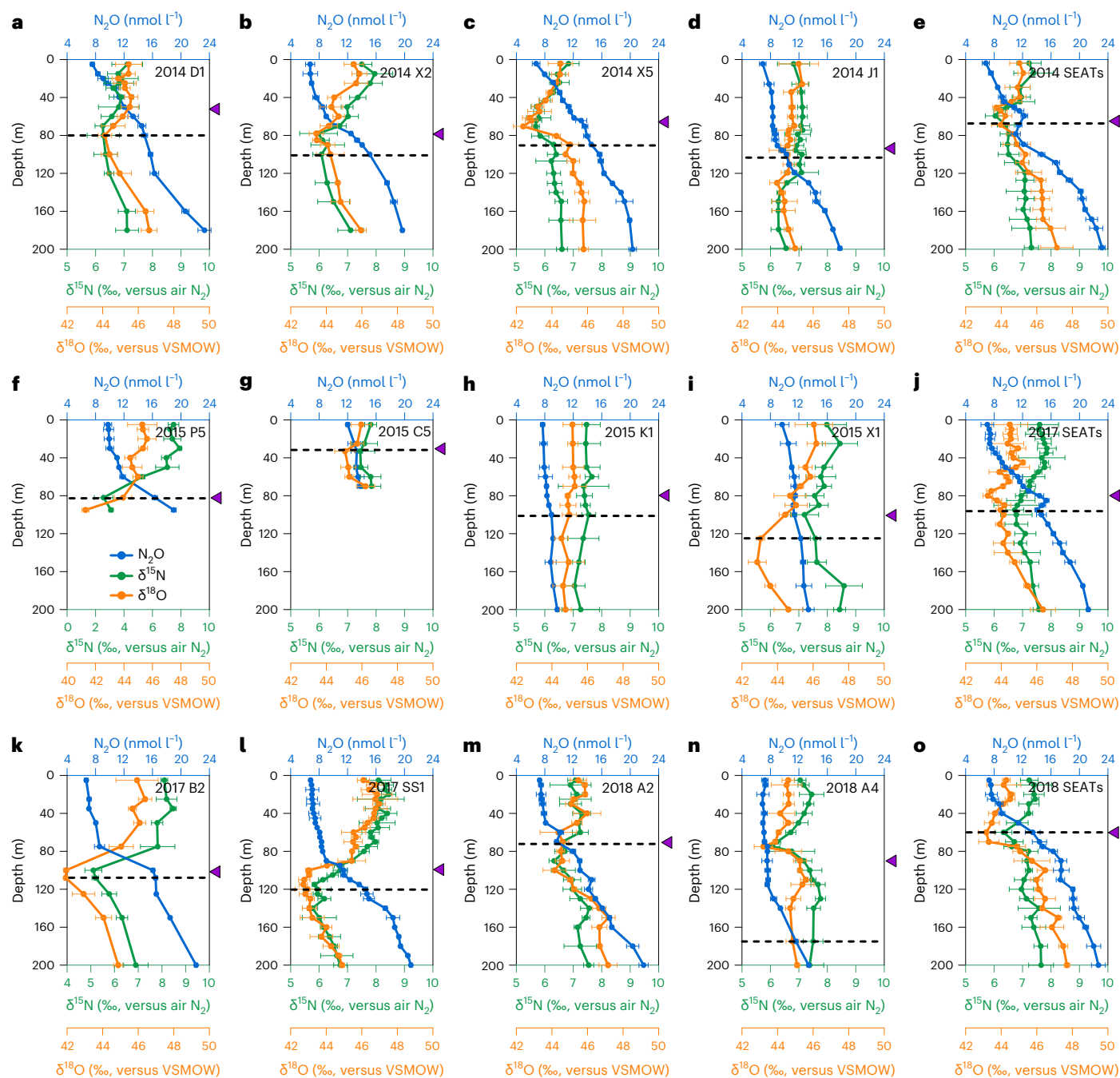
Study site (SEATs; 2014; c) and SEATs (2017; d). Blue bars mark the depth range with the distinctive N<sub>2</sub>O accumulation. Data are presented as mean values ± s.d. of triplicate sample measurement in panels c and d and are smaller than the symbols where not visible.

enzymatic details of this pathway remain mostly unresolved<sup>15,16</sup>. Moreover, sources of N<sub>2</sub>O production to the epipelagic ocean are less well studied than in the mesopelagic waters, despite the fact that remineralization of nitrogen is more intensive in the epipelagic ocean. This lack of investigation stems from the recognition that ammonia oxidation is inhibited by light<sup>17</sup> and the abundance of ammonia-oxidizing archaea is relatively low in the upper ocean<sup>13</sup>. Hence, the pathways, controls, relative contributions from various N<sub>2</sub>O sources to the N<sub>2</sub>O pool and linkages between primary productivity and N<sub>2</sub>O production in the epipelagic ocean remain unclear, hampering our ability to constrain the role of the ocean in the atmospheric N<sub>2</sub>O budget.

Nitrogen is a primary limiting nutrient to phytoplankton growth over much of the low-latitude oceans<sup>18,19</sup>. Supply of exogenous sources of nitrogen to the euphotic zone can control net productivity in these waters, with new nitrogen introduced via NO<sub>3</sub><sup>-</sup> supply from subsurface, N<sub>2</sub> fixation and nitrogen deposition<sup>20</sup>. In addition, a large fraction of primary productivity is controlled via nitrogen supplied from remineralization within the epipelagic zone, supporting regenerated production<sup>20,21</sup>. Under the assumption of steady state, new production should be quantitatively related to export of material out of the epipelagic zone into the interior waters. Thus, regenerated and export production are two competing sides of biological

productivity, where regeneration drives recycling (including N<sub>2</sub>O production), while export determines organic carbon removal to depth. Intensive organic matter remineralization not only reduces CO<sub>2</sub> sequestration efficiency, but also contributes to N<sub>2</sub>O production via nitrogen recycling. The potential counter effect of N<sub>2</sub>O emission in offsetting the radiative effect of CO<sub>2</sub> sequestration via export has been investigated in a few geoengineering and nitrogen deposition experiments<sup>22–24</sup>; however, the source of N<sub>2</sub>O, and links between biological CO<sub>2</sub> sequestration and N<sub>2</sub>O production, have not been quantified.

We hypothesized that in the vast subtropical oligotrophic oceans, where export production is inefficient<sup>3–7</sup>, the counter effect of N<sub>2</sub>O emission on carbon removal by the biological pump could be substantial. We measured rates of N<sub>2</sub>O production and carbon export in the epipelagic ocean extending from the South China Sea (SCS) into the North Pacific Subtropical Gyre (NPSG) during seven cruises conducted over eight years (Supplementary Fig. 1 and Supplementary Tables 1 and 2). We demonstrate that a large fraction of N<sub>2</sub>O in the surface ocean is locally produced in the epipelagic waters via both oxidative and reductive pathways. Furthermore, the magnitude of this shallow N<sub>2</sub>O source appears to covary spatially with biological productivity, offsetting greenhouse warming mitigation achieved by carbon export.



**Fig. 2 | Depth profiles of concentration and  $\delta^{15}\text{N}$  and  $\delta^{18}\text{O}$  isotopes of  $\text{N}_2\text{O}$  in the study area. a–e**,  $\text{N}_2\text{O}$  concentration (blue dots),  $\delta^{15}\text{N}$  (green dots) and  $\delta^{18}\text{O}$  (orange dots) at stations D1 (a), X2 (b), X5 (c), J1 (d) and SEATs (e) in the SCS in 2014. f–i, Stations P5 (f), C5 (g), K1 (h) and X1 (i) in the NWP in 2015. j–l, Stations SEATs (j), B2 (k) and SS1 (l) in the SCS in 2017. m–o, Stations A2 (m), A4 (n) and

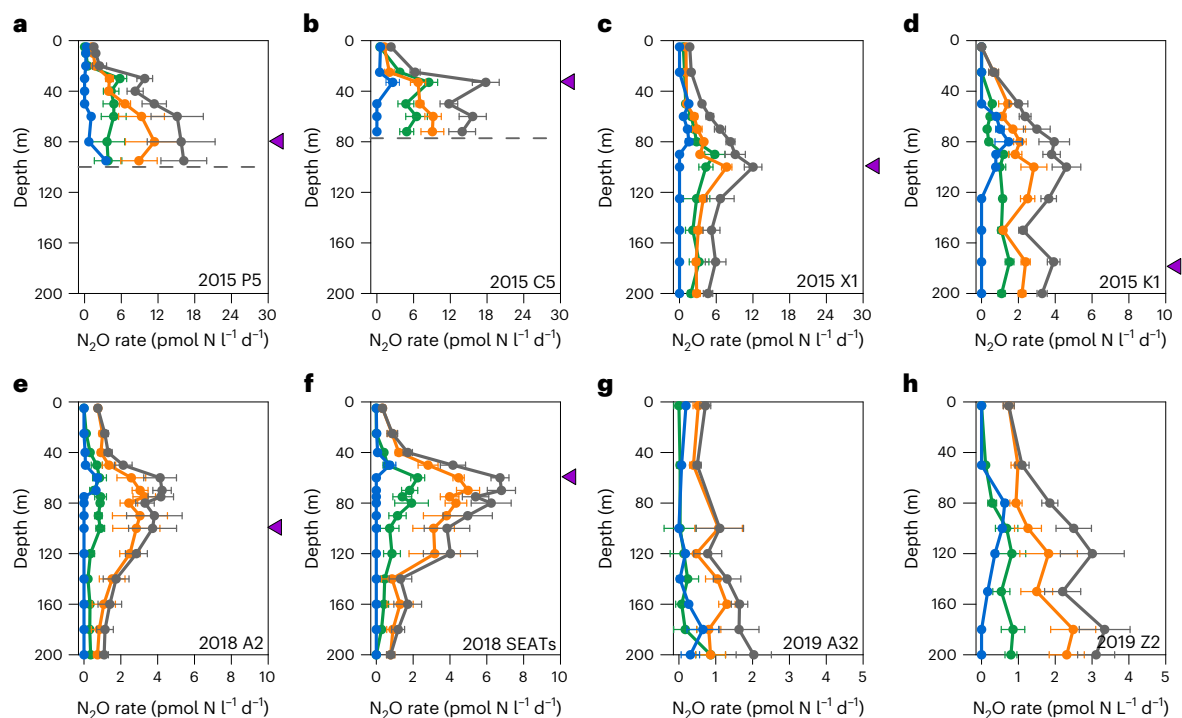
SEATs (o) in the SCS in 2018. The dashed lines mark the depth with  $\text{NO}_3^-/\text{Si}$  maximum and purple triangles mark the depth of the PNM. Data are presented as mean values  $\pm$  s.d. of triplicate sample measurement and are smaller than the symbols where not visible. Note that P5 and C5 are shelf stations.

Incorporating  $\text{N}_2\text{O}$  production that accompanies organic matter remineralization is necessary to quantify net climatic consequences associated with the marine biological pump.

### Low export efficiency in the oligotrophic subtropical ocean

The epipelagic waters at all study sites exhibited characteristics typical for the thermally stratified SCS and the NPSG, with high oxygen and low nutrient concentrations throughout the shallow mixed layer (Extended Data Figs. 1 and 2 and Supplementary Table 3).  $\text{NH}_4^+$

concentrations were persistently low (mean  $\pm$  s.d.,  $24.4 \pm 23.6 \text{ nmol l}^{-1}$  in the SCS and  $21.0 \pm 24.3 \text{ nmol l}^{-1}$  in the NPSG) with occasional peaks of  $50\text{--}200 \text{ nmol l}^{-1}$  below the mixed layer. By contrast,  $\text{NO}_2^-$  concentrations consistently showed a primary nitrite maximum (PNM) slightly below the deep chlorophyll maximum (DCM). A prominent maximum in the  $\text{NO}_3^-$  to Si concentration ratio was also frequently observed in the vicinity of the PNM layer. Particulate organic carbon (POC) and particulate nitrogen (PN) concentrations generally decreased with depth, and concentrations were greater in the shelf region than the open ocean (Extended Data Fig. 3).



**Fig. 3 | Depth profiles of  $\text{N}_2\text{O}$  production rates from multiple isotope labelling incubations.** **a–d**, Stations P5 (**a**), CS (**b**), X1 (**c**) and K1 (**d**) in the NWP in 2015. **e, f**, Stations A2 (**e**) and SEATs (**f**) in the SCS in 2018. **g, h**, Stations A32 (**g**) and Z2 (**h**) in the NPSG in 2019. The figure shows  $\text{N}_2\text{O}$  production rates from  $\text{NH}_4^+$  (green dots),  $\text{NO}_2^-$  (orange dots),  $\text{NO}_3^-$  (blue dots) and the total  $\text{N}_2\text{O}$

production rate (grey dots). Purple triangles on the right y axis mark the depth with lowest  $\delta^{15}\text{N}-\text{N}_2\text{O}$  in the investigated depth profile. The dashed lines in **a** and **b** mark the bottom depth of the shelf stations. Data are presented as mean rates  $\pm$  s.d. of triplicate sample incubation in the SCS and NPSG, and duplicates in the 2015 NWP cruise, and are smaller than the symbols where not visible.

Rates of primary production in the nutrient-depleted waters were consistently low ( $<1 \mu\text{mol C l}^{-1} \text{d}^{-1}$ ) and decreased with depth at both the SCS and the NPSG (Extended Data Fig. 4). Depth-integrated (0–125 m) primary production ranged from  $10.4 \pm 0.4$  to  $44.1 \pm 3.0 \text{ mmol C m}^{-2} \text{d}^{-1}$ , similar to long-term observations at the NPSG Station ALOHA and the BATS station in the Atlantic<sup>6,7</sup>. Carbon export at 200 m in the SCS and NPSG stations ranged from  $0.3 \pm 0.1$  to  $2.0 \pm 0.4 \text{ mmol C m}^{-2} \text{d}^{-1}$ , with low export ratios (averaging  $3.9 \pm 2.3\%$ ; Table 1 and Supplementary Discussion 1). Such results are typical of the subtropical oceans, where most primary production undergoes rapid remineralization, fuelling intensive and rapid nitrogen recycling in the epipelagic zone<sup>7,25</sup>.

### A large shallow source of $\text{N}_2\text{O}$ in the epipelagic zone

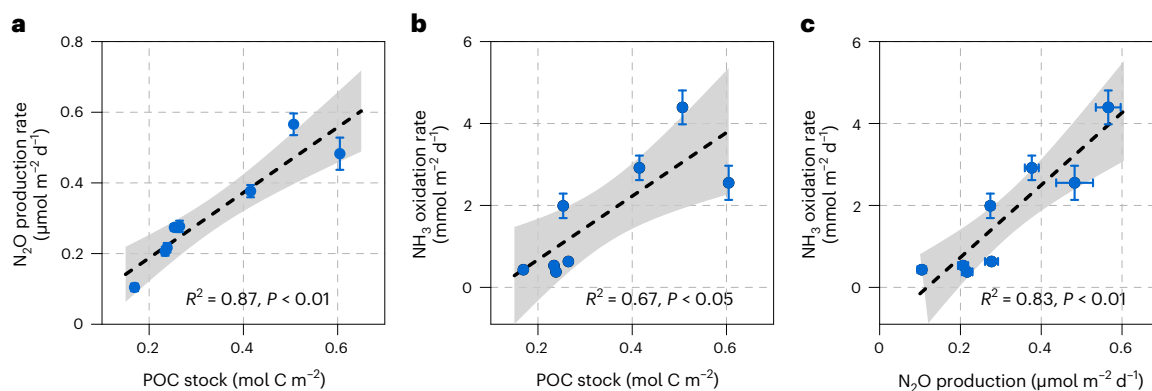
$\text{N}_2\text{O}$  concentrations in the near-surface waters were consistently near or in excess of the air-saturation state. The mean saturation value of  $\text{N}_2\text{O}$  in the coastal and shelf region ( $121 \pm 6\%$ ) was slightly higher than the open ocean ( $110 \pm 6\%$ ). The corresponding air–sea  $\text{N}_2\text{O}$  flux in the shelf and open ocean stations during the observation period was  $2.8 \pm 0.9 \mu\text{mol m}^{-2} \text{d}^{-1}$  and  $1.1 \pm 0.6 \mu\text{mol m}^{-2} \text{d}^{-1}$ , respectively, demonstrating that both regions were sources of  $\text{N}_2\text{O}$  to the atmosphere (Supplementary Table 4), in agreement with previous observations in the subtropical oceans<sup>26–28</sup>.

Our high-resolution vertical profile sampling of the epipelagic waters provided insights into the vertical variations in  $\text{N}_2\text{O}$  concentrations.  $\text{N}_2\text{O}$  concentrations and the resulting air-saturation states generally increased with depth. Distinctive  $\text{N}_2\text{O}$  concentration peaks, which deviate from simple vertical mixing, were observed at four stations (Fig. 1 and Extended Data Fig. 5). At these sites, peaks in  $\text{N}_2\text{O}$  concentrations occurred within narrow depth intervals of 10–20 m and could be easily missed by coarser vertical sampling resolution.

The location of these  $\text{N}_2\text{O}$  peaks varied in depth, temperature, salinity and density (Extended Data Fig. 6), but consistently overlapped with the PNM and  $\text{NO}_3^-/\text{Si}$  maximum layers (Fig. 1). The proximity of an  $\text{N}_2\text{O}$  peak to the PNM suggests a spatial coupling between  $\text{N}_2\text{O}$  accumulation and intensive nitrogen recycling at or around the PNM layer, where high ammonia oxidation rates occur<sup>29</sup>, leading to enhanced  $\text{N}_2\text{O}$  production and accumulation of  $\text{NO}_2^-$  (refs. <sup>30,31</sup>). The  $\text{NO}_3^-/\text{Si}$  maximum provides additional evidence of intensive remineralization of organic nitrogen and subsequent nitrification at this depth<sup>32,33</sup>, reinforcing the contribution of local sources to the  $\text{N}_2\text{O}$  accumulation.

High-resolution vertical profiles of  $\text{N}_2\text{O}$  stable isotopes ( $\delta^{15}\text{N}-\text{N}_2\text{O}$  and  $\delta^{18}\text{O}-\text{N}_2\text{O}$ ) at additional stations provided further evidence for near-surface in-situ production of  $\text{N}_2\text{O}$  in the subtropical ocean (Fig. 2). The dual isotopes of  $\text{N}_2\text{O}$  covaried with depth, decreasing from near equilibrium with the atmosphere in the near-surface waters (average  $\delta^{15}\text{N} = 7.4 \pm 0.4\text{‰}$ ;  $\delta^{18}\text{O} = 45.3 \pm 0.6\text{‰}$ ), to minima in both isotope ratios in the vicinity of the  $\text{NO}_3^-/\text{Si}$  maximum and the PNM (minima were  $0.4 \pm 0.2\text{‰}$  to  $4.9 \pm 0.3\text{‰}$  and  $0.5 \pm 0.4\text{‰}$  to  $4.0 \pm 0.4\text{‰}$  lower than the values in surface water for  $\delta^{15}\text{N}$  and  $\delta^{18}\text{O}$ , respectively). Below the PNM, the  $\delta^{15}\text{N}$  and  $\delta^{18}\text{O}$  values of  $\text{N}_2\text{O}$  increased with depth. These results indicate that the prominent dual isotope minima around the PNM probably do not derive from vertical mixing of surface and deep waters. Lateral advection of water with low isotopic signatures is also unlikely because the isotopic minimum layers occurred within waters of varying density and salinity (Extended Data Fig. 7). Hence, the most likely cause for the local  $\delta^{15}\text{N}$  and  $\delta^{18}\text{O}$  minima is in-situ  $\text{N}_2\text{O}$  production, as has previously been observed at Station ALOHA in the NPSG<sup>10,34</sup>. The widespread dual isotope minima in our study area (12 out of 15 stations) further suggest that shallow  $\text{N}_2\text{O}$  production is ubiquitous in the subtropical ocean. The shallowness of the feature in the SCS is noteworthy (60–120 m in the epipelagic zone versus ~300 m in the upper mesopelagic zone at ALOHA<sup>10,34</sup>). This shallow feature points to a potentially sensitive





**Fig. 4 | Correlations between depth-integrated  $\text{N}_2\text{O}$  production rates, ammonia oxidation rates and POC inventories.** **a**, Depth-integrated (0–200 m for the stations in open ocean and surface to bottom for the shelf stations)  $\text{N}_2\text{O}$  production rates versus POC inventories. **b**, Depth-integrated ammonia oxidation rate versus POC inventories. **c**, Depth-integrated ammonia oxidation rates versus  $\text{N}_2\text{O}$  production rates. The blue dots denote the measured rates and

POC inventory at stations P5, C5, K1, X1, A2, SEATs (2018), A32 and Z2. Data are presented as the depth-integrated rates  $\pm$  propagated s.d. of duplicates in the 2015 Northwest Pacific cruise and triplicates in the SCS and NPSG and are smaller than the symbols where not visible. The dashed black lines and grey shadows show linear regressions and the 95% confidence intervals, respectively.

climatic consequence of nitrogen recycling along ocean margins due to the shorter distance from the isotope minimum layer to the air–sea interface and more vigorous physical dynamics in the epipelagic ocean.

The relative contribution of in-situ  $\text{N}_2\text{O}$  production from the isotope minimum layer to air–sea flux can be constrained using an isotope mass balance model under the assumption that  $\text{N}_2\text{O}$  in the isotopic minima layer derives from a mixture of  $\text{N}_2\text{O}$  diffusing from the concentration maximum layer and locally produced  $\text{N}_2\text{O}$  (ref. <sup>10</sup>). Applied to our stations, this two-component model reveals that shallow  $\text{N}_2\text{O}$  production contributes  $41.6 \pm 21.0\%$  by using  $\delta^{15}\text{N}$  mass balance and  $31.3 \pm 11.0\%$  by using  $\delta^{18}\text{O}$  mass balance (Supplementary Discussion 2 and Supplementary Table 5), implying the shallow  $\text{N}_2\text{O}$  source is a substantial contributor to air–sea flux in the oligotrophic oceans.

### Multiple biological $\text{N}_2\text{O}$ sources in the oxygenated water

The underlying mechanisms that cause the  $\text{N}_2\text{O}$  isotope minimum are not fully resolved. Nitrification was previously considered a primary  $\text{N}_2\text{O}$  source to the well-oxygenated open ocean<sup>8–10</sup>. However, later observations of  $\delta^{18}\text{O}$ – $\text{N}_2\text{O}$  and isotope labelling incubations suggest part of the  $\text{N}_2\text{O}$  in the isotope minima may be produced through nitrifier denitrification or denitrification in particle-associated microenvironments<sup>34,35</sup>. Because  $\text{NO}_2^-$  incorporates oxygen atoms from water<sup>30</sup>, the  $\delta^{18}\text{O}$ – $\text{H}_2\text{O}$  signal of seawater is incorporated into  $\text{N}_2\text{O}$  from  $\text{NO}_2^-$ . In addition,  $\text{NO}_2^-$  at the PNM layer is usually depleted in  $^{15}\text{N}$ , that is,  $-0.39 \pm 3.45\%$  (Supplementary Table 6), which is lower than the reported  $\delta^{15}\text{N}$ – $\text{NO}_3^-$  ( $4.8 \pm 0.3\%$ ) and  $\delta^{15}\text{N}$ –PN ( $4.2 \pm 1.0\%$ ) at the base of the euphotic zone in the SCS<sup>36</sup>. These results indicate  $\text{NO}_2^-$  could be an important precursor to  $\text{N}_2\text{O}$ , contributing to the dual isotope minima observed in the PNM layer. Nevertheless, because  $\text{NO}_2^-$  can be incorporated into  $\text{N}_2\text{O}$  via denitrification, nitrifier denitrification or the hybrid pathway, the relative contribution of these potential sources cannot be determined using natural abundance data alone.

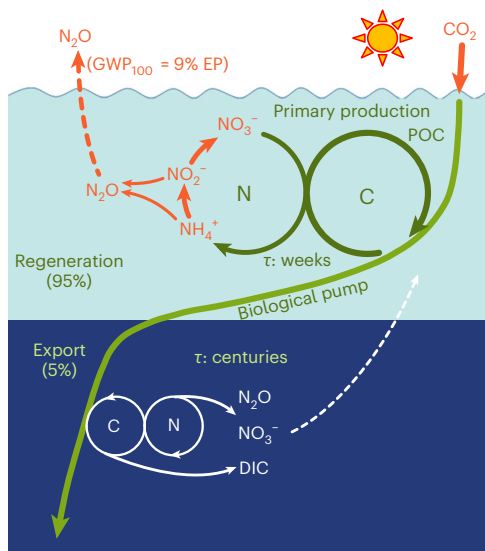
We conducted a set of  $^{15}\text{N}$  isotope tracer incubations aimed at identifying sources and quantifying their relative contributions to  $\text{N}_2\text{O}$  production. Results from these experiments show that multiple precursors contribute to  $\text{N}_2\text{O}$  production in the epipelagic ocean (Fig. 3). Notably,  $\text{N}_2\text{O}$  production was sometimes detected in the upper mixed layer even though ammonia oxidation rates were below detection limits (Extended Data Fig. 8). Gross  $\text{N}_2\text{O}$  production increased with depth to a maximum in the vicinity of the  $\text{N}_2\text{O}$  isotopic minima layer, providing additional evidence for active in-situ  $\text{N}_2\text{O}$  production. Both  $\text{NH}_4^+$  and

$\text{NO}_2^-$  seem to be involved in  $\text{N}_2\text{O}$  production, while  $\text{NO}_3^-$  reduction to  $\text{N}_2\text{O}$  was occasionally detected. Depth-integrated  $\text{N}_2\text{O}$  production rates (0–200 m) ranged from  $0.10 \pm 0.02$  to  $0.57 \pm 0.03 \mu\text{mol m}^{-2} \text{d}^{-1}$  (average  $0.28 \pm 0.04 \mu\text{mol m}^{-2} \text{d}^{-1}$ ) at the open ocean stations. These rates account for  $29.5 \pm 1.8\%$  to  $61.3 \pm 6.4\%$  (average  $40.0 \pm 7.7\%$ ) of the air–sea  $\text{N}_2\text{O}$  flux in the open ocean stations (Supplementary Table 4), demonstrating that a large proportion of air–sea  $\text{N}_2\text{O}$  flux in the subtropical ocean can be produced locally in the epipelagic zone.

Production of both  $^{45}\text{N}_2\text{O}$  and  $^{46}\text{N}_2\text{O}$  was detected, with  $^{45}\text{N}_2\text{O}$  frequently comprising a higher fraction of labelled  $\text{N}_2\text{O}$  than  $^{46}\text{N}_2\text{O}$ . The fraction of measured  $^{45}\text{N}_2\text{O}$  frequently exceeded that predicted from the binomial distribution. Excess  $^{45}\text{N}_2\text{O}$  has also been observed in several studies conducted in the mid-latitude North Atlantic<sup>37</sup>, the western North Pacific<sup>38</sup> and the eastern tropical South Pacific<sup>39,40</sup> and has been interpreted as evidence of hybrid  $\text{N}_2\text{O}$  production. However, isotope dilution of the tracer substrates, and ammonia oxidation coupled to  $\text{NO}_2^-$  reduction during the incubation, would also cause deviation of the measured  $^{45}\text{N}_2\text{O}$ : $^{46}\text{N}_2\text{O}$  from the predicted ratio, and care should be taken in interpreting the observed excess  $^{45}\text{N}_2\text{O}$  (Supplementary Discussion 3). The presence of  $^{46}\text{N}_2\text{O}$  in both  $^{15}\text{NO}_2^-$  and  $^{15}\text{NO}_3^-$  labelling incubations suggests production of  $\text{N}_2\text{O}$  via nitrifier denitrification and/or denitrification in micro-anoxic niches in the oxygenated ocean<sup>11,12</sup>. As nitrification is widely used as a key component for model parameterization to estimate  $\text{N}_2\text{O}$  production in the oxygenated ocean<sup>8,9,41</sup>, our results strongly support the contribution of multiple precursors and pathways of  $\text{N}_2\text{O}$  in the epipelagic ocean that need to be considered in biogeochemical models aiming to estimate marine  $\text{N}_2\text{O}$  sources and air–sea flux.

### $\text{N}_2\text{O}$ production offsets $\text{CO}_2$ removal by the biological pump

The export of organic matter to the ocean’s interior through the marine biological pump is a primary control on the oceanic  $\text{CO}_2$  sink on long-term timescales<sup>3–5</sup>. However, the magnitude of the marine biological pump depends on complex interactions, including those that alter the vertical length-scale of organic matter remineralization, altering the timescales over which carbon is sequestered<sup>42</sup>. Rapid (days to weeks) remineralization of organic matter and concomitant nitrogen recycling in the epipelagic zone sustains a large fraction of biological productivity throughout the subtropical oceans<sup>43–45</sup>. Our results highlight that this process also promotes  $\text{N}_2\text{O}$  production. We observed that rates of  $\text{N}_2\text{O}$  production and ammonia oxidation were significantly positively



**Fig. 5 | Schematic of the proposed linkage between surface  $\text{N}_2\text{O}$  production and  $\text{CO}_2$  removal by the biological pump.** Carbon (C) and nitrogen (N) undergo rapid cycling in the epipelagic ocean and slow cycling in the ocean's interior. During the operation of biological pump processes, only a small fraction of newly produced organic matter (in our study, only around 5% of primary production) was exported to depth. This export (EP) drives slow recycling of C and N, during which  $\text{N}_2\text{O}$  and  $\text{CO}_2$  are produced and accumulate at longer timescales (the residence time ( $\tau$ ): centuries to millennia) before exchanging with the atmosphere. In contrast, most of the newly produced organic matter is rapidly remineralized ( $\tau$ : days to weeks) in the epipelagic ocean to drive intense recycling of C and N, during which  $\text{N}_2\text{O}$  is produced through both oxidative and reductive pathways and emitted to the atmosphere. This more rapid recycling in the epipelagic waters can offset a substantial part (6–27%, median 9%) of the decreased radiative forcing ( $\text{GWP}_{100}$ ) due to biological  $\text{CO}_2$  removal to depth in our study.

correlated (Fig. 4a), highlighting the covariance between  $\text{N}_2\text{O}$  production and nitrogen regeneration in the epipelagic zone. Moreover, both the integrated ammonia oxidation rates and  $\text{N}_2\text{O}$  production rates were also significantly correlated with the POC and PN inventories (Fig. 4b,c and Extended Data Fig. 9), implying the strength of nitrogen recycling and  $\text{N}_2\text{O}$  production scales with the availability of organic nitrogen. Particles are known hotspots of microbial metabolism and can provide a source of organic and inorganic substrates to the surrounding seawater. The microenvironment formed in the particle also favours various nitrogen transformation pathways contributing to  $\text{N}_2\text{O}$  production<sup>11</sup>. In our study, larger POC and PN stocks, presumably sustained by efficient epipelagic recycling, seem to promote  $\text{N}_2\text{O}$  production.

Carbon and nitrogen cycling are intimately coupled to each other because both elements are required by all organisms. Although the regulation of carbon cycling by nitrogen supply has been extensively studied, we provide new perspectives on potential climatic impacts associated with nitrogen recycling and  $\text{N}_2\text{O}$  production in the epipelagic ocean. Such shallow  $\text{N}_2\text{O}$  production is of particular importance in the vast subtropical oligotrophic oceans, where export efficiencies are low and nutrient recycling is rapid. For example, the average export ratio in our study was  $3.9 \pm 2.3\%$  (Table 1), suggesting >95% of primary production was remineralized in the epipelagic zone. Our observations indicate that nitrogen recycling promotes production of  $\text{N}_2\text{O}$  via multiple pathways in these remineralization-intensive systems. In comparing the potential offset in radiative warming due to  $\text{N}_2\text{O}$  production relative to carbon export, we assumed a 100-year time horizon of global warming potential ( $\text{GWP}_{100}$ ) for both processes (where 1 mol  $\text{N}_2\text{O}$  would be equivalent to 300 mol  $\text{CO}_2$  in radiative energy)<sup>1</sup>. We estimate the integrated  $\text{N}_2\text{O}$  production rate associated with nitrogen recycling

would be equivalent to offsetting  $5.6 \pm 0.6\%$  to  $27.2 \pm 6.7\%$  (median value 8.8%) of the greenhouse gas mitigation capacity supported by carbon export measured at the SCS and the NPSG stations (Fig. 5 and Table 1). However, there are uncertainties associated with this estimate, variation in time and length-scales of particle remineralization and water mass ventilation would alter these radiative warming offset estimates. Therefore, our estimates would probably fall at the lower end of the potential offset attributable to  $\text{N}_2\text{O}$  production in this region. For example, ventilation times between 200 and 300 m in this region average  $32 \pm 5$  years, with that age increasing to 50 years at 400 m and 100 years at 500 m<sup>46</sup>. Assuming the vertical attenuation of sinking particulate matter follows a power-law function<sup>47</sup>, we estimate  $-57 \pm 5\%$  of the measured exported carbon would be remineralized above 500 m and could exchange with the atmosphere in <100 years, leading to less  $\text{CO}_2$  sequestration and a higher  $\text{N}_2\text{O}$  offset value (Supplementary Discussion 4). This offset of the effectiveness of the  $\text{CO}_2$  sink, attributable to a largely overlooked epipelagic  $\text{N}_2\text{O}$  source, requires re-examination of the warming mitigation capacity of the marine biological pump.

Ongoing warming of the ocean and atmosphere may lead to a decline in export efficiency and decreased length-scale of remineralization due to intensified upper-ocean stratification and shifting of phytoplankton communities towards smaller cells<sup>48</sup>. Together with increased temperatures, these dynamics may enhance organic matter recycling in the epipelagic ocean<sup>49,50</sup>, with concomitant impacts of  $\text{N}_2\text{O}$  production<sup>41</sup>. Our study suggests enhancement of surface  $\text{N}_2\text{O}$  production, through intensified organic matter remineralization, could further exacerbate warming of the climatic system through decreased export and greater  $\text{N}_2\text{O}$  production. Our results were derived from a limited number of stations at one time and cannot be directly extrapolated to the large spatial–temporal variation in both carbon export and nitrogen regeneration in the ocean<sup>3,6,7</sup>. Nevertheless, our findings show that active  $\text{N}_2\text{O}$  production, driven by intense organic matter recycling in the epipelagic ocean, can offset a considerable fraction of the benefit of radiative forcing achieved by  $\text{CO}_2$  sequestration via the marine biological pump. Future work should investigate and compare the  $\text{N}_2\text{O}/\text{CO}_2$  offset between systems with different export efficiencies. A better integrated assessment should take  $\text{N}_2\text{O}$  generation into account for understanding the climatic impact of the marine biological pump in order to devise the best greenhouse gas mitigation strategy.

## Online content

Any methods, additional references, Nature Portfolio reporting summaries, source data, extended data, supplementary information, acknowledgements, peer review information; details of author contributions and competing interests; and statements of data and code availability are available at <https://doi.org/10.1038/s41561-022-01090-2>.

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## Methods

### Field sampling and on-deck incubations

Samples were collected from high-resolution vertical profiles during four research cruises conducted during 2012 to 2015 to the SCS and the subtropical northwest Pacific aboard the RV *Dongfanghong II*; additional sampling occurred during three cruises in 2017 to 2019 in the SCS and the NPSG aboard the RV *Tan Kah Kee*. Stations spanned a wide range of hydrographic conditions and biological activities from the coastal shelf to open ocean (Supplementary Fig. 1 and Supplementary Table 1).

Temperature, salinity, depth and fluorescence concentrations were measured using a Seabird SBE 911 CTD sensor package equipped with fluorometer. Photosynthetically active radiation (PAR) was measured using PAR sensors (LI-COR Biosciences, LI-193 on RV *Dongfanghong II* and Biospherical QCP-2300L-HP on RV *Tan Kah Kee*). Discrete seawater samples were collected using 24 12-litre Niskin bottles mounted to the conductivity, temperature and depth (CTD) rosette. The base of the mixed layer was defined as the depth where a difference of 0.8 °C relative to the surface value was observed<sup>51</sup>. The nitracline was derived as the mid-point (average) of the steepest nitrate concentration gradient with depth<sup>51</sup>. The depth with 0.1% surface PAR was defined as the base of euphotic zone<sup>6</sup>.

On-board incubation was conducted at six stations across the shelf to the open ocean of the NPSG; at two of the stations in the SCS basin, experiments were conducted to quantify N<sub>2</sub>O production rates and nitrification rates in the epipelagic zone (0–200 m) of the open ocean stations and throughout the water column in stations sampled along the shelf.

Samples for chemical, biological and rate measurements were collected from the same casts. Triplicate 150 ml high-density polyethylene Nalgene bottles were used for nutrient collection; 120 ml glass serum bottles were used to collect samples for subsequent N<sub>2</sub>O concentration measurements in 2012 and 2013; and triplicate 250 ml glass serum bottles (Wheaton) were used for subsequent N<sub>2</sub>O concentration and isotope measurements from 2014 to 2018. Ammonia oxidation and N<sub>2</sub>O production rate incubations were conducted in 120 ml glass serum bottles. Seawater for subsequent analyses of POC and PN were collected into 4 l polycarbonate Nalgene bottles. All bottles and equipment were acid washed and rinsed with in-situ seawater at least three times prior to sample collection. During sample collection, glass sample bottles were overfilled two to three times before sealing with 20 mm butyl stopper and aluminium crimp seals (Wheaton). Samples were preserved by adding 0.1 ml to 0.2 ml saturated HgCl<sub>2</sub> and were stored at 4 °C. For POC and PN samples, 4–8 l of seawater was gently (<200 mm Hg, 26.6 kPa) filtered through a pre-combusted (450 °C for 4 h) Whatman GF/F filter (25 mm diameter). After filtration, the filters were folded and wrapped in pre-combusted (450 °C for 4 h) aluminium foil and stored at –80 °C.

A comprehensive set of incubations was carried out on-board to determine rates of nitrification and N<sub>2</sub>O production using <sup>15</sup>NH<sub>4</sub><sup>+</sup>, <sup>15</sup>NO<sub>2</sub><sup>-</sup> and <sup>15</sup>NO<sub>3</sub><sup>-</sup> tracers (Supplementary Table 2). All incubations were conducted in the dark at near in-situ temperatures (±2 °C). On the 2015 cruise, 0.2 ml of tracer was injected into each bottle to obtain final concentrations of <sup>15</sup>NH<sub>4</sub><sup>+</sup> and <sup>15</sup>NO<sub>2</sub><sup>-</sup> of 300 nmol l<sup>-1</sup>. For the <sup>15</sup>NO<sub>3</sub><sup>-</sup> tracer, two different stocks were used to obtain final concentrations of <sup>15</sup>NO<sub>3</sub><sup>-</sup> of 300 nmol l<sup>-1</sup> for samples above the DCM and 1,000 nmol l<sup>-1</sup> for samples below the DCM. On the 2018 SCS cruise and 2019 NPSG cruise, 0.2 ml of mixed tracer was added to each bottle to obtain enrichment of 500 nmol l<sup>-1</sup> of <sup>15</sup>NH<sub>4</sub><sup>+</sup> + <sup>14</sup>NO<sub>2</sub><sup>-</sup> or <sup>14</sup>NH<sub>4</sub><sup>+</sup> + <sup>15</sup>NO<sub>2</sub><sup>-</sup>. For <sup>15</sup>NO<sub>3</sub><sup>-</sup> tracer, two different stocks were used to get a final enrichment of 500 nmol l<sup>-1</sup> of <sup>15</sup>NO<sub>3</sub><sup>-</sup> for samples above the DCM and 1,000 nmol l<sup>-1</sup> of <sup>15</sup>NO<sub>3</sub><sup>-</sup> below the DCM in both the 2018 SCS cruise and 2019 NPSG cruise. The final tracer concentrations were frequently higher than the in-situ substrate concentrations, accounting for 89 ± 12%, 85 ± 15% and 42 ± 30% of the final substrate pool in the <sup>15</sup>NH<sub>4</sub><sup>+</sup>, <sup>15</sup>NO<sub>2</sub><sup>-</sup> and <sup>15</sup>NO<sub>3</sub><sup>-</sup> labelling incubations, respectively. <sup>15</sup>NH<sub>4</sub><sup>+</sup> labelling incubations were conducted for deriving both N<sub>2</sub>O production and nitrification rates. Immediately after the <sup>15</sup>NH<sub>4</sub><sup>+</sup> tracer injection, 10 ml

of sample was pushed out by pure N<sub>2</sub> and then filtered through a 0.2 µm syringe filter to represent the initial condition (t<sub>0</sub>) for the nitrification incubations. The remaining water was preserved with 0.1 ml saturated HgCl<sub>2</sub>. The remaining bottles were incubated in the dark at near in-situ temperature. At each timepoint, 10 ml of water was sampled and then filtered for subsequent nitrification rate measurements, and the remaining water was preserved using HgCl<sub>2</sub> for subsequent determinations of N<sub>2</sub>O. The filtrate was stored at –20 °C for subsequent analyses. For <sup>15</sup>NO<sub>2</sub><sup>-</sup> and <sup>15</sup>NO<sub>3</sub><sup>-</sup> incubations, the same procedures were used for on-board incubation, except that the incubation was terminated by adding 0.1 ml HgCl<sub>2</sub> without replacing by N<sub>2</sub>. Primary production rate was also measured in selected stations on 2015, 2018 and 2019 cruises using H<sup>13</sup>CO<sub>3</sub><sup>-</sup> tracer (99 atom% <sup>13</sup>C, Cambridge Isotope Laboratories), and the final concentration of H<sup>13</sup>CO<sub>3</sub><sup>-</sup> was 100 µmol l<sup>-1</sup>, accounting for ~5% of the substrate pool. On-deck incubation (duplicates) was performed in 4 l polycarbonate Nalgene bottles for 24 h. Light conditions of the incubators were manipulated by neutral density filter. Seawater was gently (<200 mm Hg, 26.6 kPa) filtered through a pre-combusted (450 °C for 4 h) Whatman GF/F filter (25 mm diameter) and stored at –80 °C.

### Nutrient, POC and PN measurements

NH<sub>4</sub><sup>+</sup> concentrations were measured on-board the research vessels using a fluorometric method with detection limit of 1.2 nmol l<sup>-1</sup> and precision of ±3.5%<sup>52</sup>. Nutrient concentrations below the nitracline were measured using a four-channel Continuous Flow Technicon AA3 Auto-Analyzer. The detection limits for NO<sub>x</sub> (NO<sub>3</sub><sup>-</sup> + NO<sub>2</sub><sup>-</sup>) and Si(OH)<sub>4</sub> were 0.03 µmol l<sup>-1</sup> and 0.05 µmol l<sup>-1</sup>, respectively, with precision better than 1% and 2.8%, respectively<sup>53</sup>. NO<sub>2</sub><sup>-</sup> and NO<sub>3</sub><sup>-</sup> concentrations above the nitracline were determined using the standard colorimetric method coupled with a Flow Injection Analysis-Liquid Waveguide Capillary Cell system (World Precision Instruments)<sup>54</sup>; the detection limit was 5 nmol l<sup>-1</sup> and precision was better than 3.1%. For POC and PN concentration measurement, the filters were freeze dried and then acidified with 1 ml of 1 N HCl solution to remove carbonates. All filters were dried at 60 °C for 48 h. The decarbonated samples were then analysed for POC and PN using an EA-IRMS (Thermo Finnigan Flash EA 2000 interfaced to a Delta V<sup>PLUS</sup> isotopic ratio mass spectrometer) system. The precision for both PN and POC concentration is <1% (ref. <sup>55</sup>).

### N<sub>2</sub>O concentration measurement

During 2012 to 2013, N<sub>2</sub>O concentrations were measured using a purge and trap system coupled with a gas chromatograph (Hewlett-Packard model 6890 equipped with a micro-electron capture detector). Calibration of N<sub>2</sub>O concentrations was determined from peak areas with standard gases of 1.0–5.0 ppmv N<sub>2</sub>O/N<sub>2</sub> (Research Institute of China National Standard Materials), which were run at six-sample intervals. The precision of this method was estimated to be better than ±5% (ref. <sup>56</sup>). Beginning in 2014, N<sub>2</sub>O concentrations were also derived from ion peak area (m/z = 44) during isotope analysis using the gas chromatography-isotope ratio mass spectrometry (GC-IRMS) system (see below). The two methods yielded comparable results; thus, N<sub>2</sub>O concentrations are shown as the mean value from these independent methods.

### <sup>234</sup>Th measurement

The thorium-deficit method was used to estimate export production. Total <sup>234</sup>Th samples were processed using a manganese oxide co-precipitation technique<sup>57</sup>. Briefly, total <sup>234</sup>Th in the seawater was co-precipitated with MnO<sub>2</sub> particles and the resulting particles were collected on a 25 mm, 1.0 µm quartz micro-filter (QMA). Suspended particles in the seawater were also analysed for <sup>234</sup>Th; for these samples, ~8 l water was filtered onto a QMA filter. All total and particulate <sup>234</sup>Th samples were beta counted on a gas flow proportional low-level Risø beta-counter for 16 h until total counts >2,500. A second counting was carried out after >150 days for background correction. The recovery for <sup>234</sup>Th was monitored by <sup>230</sup>Th spike addition in the seawater and

quantified by an alpha-counter with addition of a  $^{228}\text{Th}$  internal standard<sup>58</sup>. The recoveries of  $^{234}\text{Th}$  were better than 90%.  $^{238}\text{U}$  (dpm  $\text{l}^{-1}$ ) was calculated from the linear relationship of  $^{238}\text{U}$  with salinity<sup>59</sup>.

### Isotopic analyses of $\text{NO}_x^-$ and $\text{N}_2\text{O}$

$\delta^{15}\text{N}$  of  $\text{NO}_x^-$  samples for nitrification rate were determined using the denitrifier method<sup>60,61</sup>. Briefly,  $\text{NO}_x^-$  was quantitatively converted to  $\text{N}_2\text{O}$  using the bacterial strain *Pseudomonas aureofaciens*. The evolved  $\text{N}_2\text{O}$  was then introduced to the GC-IRMS (Delta V<sup>PLUS</sup> isotopic ratio mass spectrometer) through an online  $\text{N}_2\text{O}$  cryogenic extraction and purification system.  $\delta^{15}\text{N}$  of  $\text{NO}_x^-$  values were calibrated against nitrate isotope standards USGS 34, IAEA N3 and USGS 32, which were run before, after and at ten-sample intervals. Accuracy was better than  $\pm 0.2\%$  according to analyses of these standards at an injection level of 20 nmol N. For samples with  $\text{NO}_x^-$  concentrations lower than  $0.5 \mu\text{mol l}^{-1}$ , 1 ml of  $5 \mu\text{mol l}^{-1}$  of in-house  $\text{NO}_3^-$  standard was added as carrier to 9 ml of sample, and the isotopic composition of the sample was then calculated from the measured composition of the mixture and the known in-house standard via mass conservation.

Concentrations and isotopes of  $\text{N}_2\text{O}$  were measured using a modified GC-IRMS with large volume purge and trap system<sup>62</sup>. Briefly, two needles were used for sample transfer and He pressurization, and the sample was transferred into a sparging flask (Pyrex) using ultra-high-purity He (>99.999%) and purged with He. For a 250 ml bottle, the sample was purged for 60 min at a flow rate of  $50 \text{ ml min}^{-1}$ , and for a 120 ml bottle, the purge time was 30 min. The extracted gases were passed through an ethanol trap with dry ice and a chemical trap filled with magnesium perchlorate and Ascarite to remove  $\text{H}_2\text{O}$  and  $\text{CO}_2$ .  $\text{N}_2\text{O}$  was trapped by liquid nitrogen twice for purification and concentration and then injected into the GC-IRMS with He as carrier gas.  $\text{N}_2\text{O}$  concentrations were determined by ion peak area ( $m/z = 44$ ), and calibration of  $\text{N}_2\text{O}$  concentration was calculated from ion peak areas ( $m/z = 44$ ) with standard gases of 199.6 and 501.0 ppmv  $\text{N}_2\text{O}/\text{He}$ , which were run at ten-sample intervals. The serum bottle was weighed before and after transfer to calculate the amount of water transferred. The precision of this method was estimated to be better than  $\pm 3\%$  (ref. <sup>62</sup>).  $\delta^{15}\text{N}$  and  $\delta^{18}\text{O}$  were calibrated against two reference tanks (R1: 199.6 ppmv  $\text{N}_2\text{O}/\text{He}$ ,  $\delta^{15}\text{N} = -3.2 \pm 0.1\%$  relative to air  $\text{N}_2$ ,  $\delta^{18}\text{O} = 36.6 \pm 0.1\%$  relative to Vienna Standard Mean Ocean Water; R2: 501.0 ppmv  $\text{N}_2\text{O}/\text{He}$ ,  $\delta^{15}\text{N} = -1.6 \pm 0.1\%$ ,  $\delta^{18}\text{O} = 36.6 \pm 0.3\%$ ), which were measured in the Casciotti lab at Stanford University. The precision of  $\delta^{15}\text{N}$  and  $\delta^{18}\text{O}$  measurements with 2 nmol  $\text{N}_2\text{O}$  reference gas was better than  $0.3\%$  and  $0.4\%$ , respectively ( $n = 20$ ).

### Surface $\text{N}_2\text{O}$ saturation and air–sea flux

Surface  $\text{N}_2\text{O}$  saturation was calculated using equation (1):

$$R = \frac{C_{\text{obs}}}{C_{\text{eq}}} \times 100 \quad (1)$$

where  $R$  (%) is the saturation of surface  $\text{N}_2\text{O}$ ;  $C_{\text{obs}}$  represents  $\text{N}_2\text{O}$  concentration at 5 m depth;  $C_{\text{eq}}$  is the expected equilibrium concentration, which is computed using Henry's law<sup>63</sup>; and the solubility depends on temperature and salinity<sup>64</sup>. The air  $\text{N}_2\text{O}$  concentration is the average atmospheric  $\text{N}_2\text{O}$  concentration at Mauna Loa of the sampling year (NOAA/ESRL programme).

Air–sea  $\text{N}_2\text{O}$  flux was computed using equations (2) and (3):

$$F = k \times (C_{\text{obs}} - C_{\text{eq}}) \quad (2)$$

$$k = 0.251 \times u^2 \times \left(\frac{\text{Sc}}{660}\right)^{-0.5} \quad (3)$$

where  $F$  ( $\mu\text{mol m}^{-2} \text{d}^{-1}$ ) is air–sea flux of  $\text{N}_2\text{O}$ ;  $k$  ( $\text{cm h}^{-1}$ ) is the gas transfer velocity depending on wind and water temperature;  $u$  is daily mean

wind speed at 10 m above sea surface during the cruise, as measured by the on-board meteorological station; and  $\text{Sc}$  is the Schmidt number calculated from temperature<sup>64</sup>.

### Estimation of the fraction of $\text{N}_2\text{O}$ source derived from the isotope minimum layer

A two-endmember mixing model of isotopically enriched  $\text{N}_2\text{O}$  mixing upward from the  $\text{N}_2\text{O}$  concentration maximum layer and isotopically depleted  $\text{N}_2\text{O}$  produced at the isotope minima layer was used to calculate the fraction of  $\text{N}_2\text{O}$  contributed by shallow in-situ production using equation (4)<sup>10</sup>:

$$\delta_{\text{shallow}} = \frac{\delta_{\text{total}} - (1-f)\delta_{\text{deep}}}{f} \quad (4)$$

where  $\delta_{\text{total}}$  is the lowest measured isotopic value of  $\text{N}_2\text{O}$  at the isotope minimum.  $\delta_{\text{deep}}$  is the isotopic signature of  $\text{N}_2\text{O}$  mixing upward from deep layers; here, we use the  $\text{N}_2\text{O}$  concentration maximum layer as an endmember; the measured  $\delta^{15}\text{N}$  was  $9.52 \pm 0.28\%$  and the  $\delta^{18}\text{O}$  was  $52.25 \pm 0.74\%$  in our study sites (Extended Data Fig. 10).  $\delta_{\text{shallow}}$  is the isotopic value of the in-situ source in the isotope minimum layer, which is unknown.  $f$  is the fraction of  $\text{N}_2\text{O}$  contributed from the shallow source to the isotope minimum layer, with the remainder equal to that diffusing upward from the  $[\text{N}_2\text{O}]$  maximum. The lower limit of  $f$  could be constrained by assuming  $\delta_{\text{shallow}}$  was represented by the lowest value in an existing database from the North Pacific, and the  $\delta^{15}\text{N}$  and  $\delta^{18}\text{O}$  was  $1.8\%$  and  $24.5\%$ , respectively<sup>65</sup>.

### Nitrification and $\text{N}_2\text{O}$ production rate calculation

Rates of nitrification were determined based on the accumulation of  $^{15}\text{N}$  in the product pool relative to the initial  $^{15}\text{N}$  signature of that pool. Rates were computed based on equation (5):

$$R_{\text{NR}} = \frac{1}{t} \times \frac{C_{\text{NO}_x^-} \times (n_t - n_0)}{f^{A5}} \times 24 \quad (5)$$

where  $R_{\text{NR}}$  is the bulk nitrification rate for all substrates following  $^{15}\text{NH}_4^+$  enrichment ( $\text{nmol N l}^{-1} \text{d}^{-1}$ ).  $C_{\text{NO}_x^-}$  is the product concentration at the beginning of incubation ( $\text{nmol N l}^{-1}$ ).  $f^{A5}$  is the atom%  $^{15}\text{N}$  of the  $\text{NH}_4^+$  pool at the beginning of incubation (the fraction of  $^{15}\text{N}-\text{NH}_4^+$  in the gross  $\text{NH}_4^+$  pool after tracer enrichment), and  $n_t$  and  $n_0$  are the atom%  $^{15}\text{N}$  of the product pool ( $\text{NO}_2^- + \text{NO}_3^-$ ) at the ending and beginning of incubation (%), respectively.  $t$  is the duration of incubation (h). This equation quantifies the transformation rate including the concentration due to the tracer addition (that is, ambient substrate + tracer) and thus represents a potential reaction rate.

Rates of  $\text{N}_2\text{O}$  production from a particular labelled substrate (for example,  $^{15}\text{NH}_4^+$ ) were quantified as the increase in mass 44, 45 and 46 from  $\text{NH}_4^+$  during an incubation. In our calculation,  $^{15}\text{N}-\text{N}_2\text{O}$  production during  $^{15}\text{NH}_4^+$  incubation was obtained from the increase of 45 and 46, and the  $^{14}\text{N}-\text{N}_2\text{O}$  production from  $^{14}\text{NH}_4^+$  was then derived based on the atom fraction of  $^{14}\text{N}$  and  $^{15}\text{N}$  of the substrate pool. During the incubations, the tracer substrate is enriched in  $^{15}\text{N}$ ; thus, we assume the accumulation of  $^{45}\text{N}_2\text{O}$  is mainly contributed by  $^{15}\text{N}^{14}\text{N}^{16}\text{O}$  and  $^{14}\text{N}^{15}\text{N}^{16}\text{O}$  (single labelled  $\text{N}_2\text{O}$ ) and production of  $^{14}\text{N}^{14}\text{N}^{17}\text{O}$  during the incubation is negligible. Similarly, the accumulation of  $^{46}\text{N}_2\text{O}$  is thus mainly contributed by  $^{15}\text{N}^{15}\text{N}^{16}\text{O}$  (double labelled  $\text{N}_2\text{O}$ ), and the production of  $^{14}\text{N}^{15}\text{N}^{17}\text{O}$  and  $^{14}\text{N}^{14}\text{N}^{18}\text{O}$  are negligible compared with the double labelled  $\text{N}_2\text{O}$ . Therefore, the rates of  $^{45}\text{N}_2\text{O}$  production and  $^{46}\text{N}_2\text{O}$  production can be derived using equations (6) and (7):

$$R_{45\text{measure}} = \frac{1}{t} \times C_{\text{N}_2\text{O}} \times (n_{t45} - n_{045}) \times 24 \quad (6)$$

$$R_{46\text{measure}} = \frac{1}{t} \times C_{\text{N}_2\text{O}} \times (n_{t46} - n_{046}) \times 24 \quad (7)$$

where  $R_{45\text{measure}}$  ( $\text{pmol N}_2\text{O l}^{-1} \text{d}^{-1}$ ) is measured production rate of  $^{45}\text{N}_2\text{O}$  according to the increase of measured  $R^{45}\text{N}_2\text{O}/^{44}\text{N}_2\text{O}$ ;  $R_{46\text{measure}}$  ( $\text{pmol N}_2\text{O l}^{-1} \text{d}^{-1}$ ) is measured production rate of  $^{46}\text{N}_2\text{O}$  according to the increase of measured  $R^{46}\text{N}_2\text{O}/^{44}\text{N}_2\text{O}$ ; and  $C_{\text{N}_2\text{O}}$  is  $\text{N}_2\text{O}$  concentration ( $\text{pmol N}_2\text{O l}^{-1}$ ) at the beginning of the incubation.  $n_{t45}$ ,  $n_{045}$ ,  $n_{t46}$  and  $n_{046}$  are the  $^{45}\text{N}_2\text{O}\%$  and  $^{46}\text{N}_2\text{O}\%$  based on  $R^{45}\text{N}_2\text{O}/^{44}\text{N}_2\text{O}$  and  $R^{46}\text{N}_2\text{O}/^{44}\text{N}_2\text{O}$  at the ending and beginning of incubation (%), respectively.  $t$  is the duration of incubation (h), which is converted to units of days (d) by multiplying by  $24 \text{ h d}^{-1}$ .

The production of  $^{15}\text{N}-\text{N}_2\text{O}$  and  $^{14}\text{N}-\text{N}_2\text{O}$  from a single substrate (for example,  $\text{NH}_4^+$ ) can be then derived using equations (8) and (9):

$$R_{^{15}\text{N}-\text{N}_2\text{O}} = R_{45\text{measure}} + R_{46\text{measure}} \times 2 \quad (8)$$

$$R_{^{14}\text{N}-\text{N}_2\text{O}} = R_{^{15}\text{N}-\text{N}_2\text{O}} \times \frac{f_{14}}{f_{15}} \quad (9)$$

where  $R_{^{15}\text{N}-\text{N}_2\text{O}}$  ( $\text{pmol N l}^{-1} \text{d}^{-1}$ ) is the measured production of  $^{15}\text{N}-\text{N}_2\text{O}$  from the labelled substrate (note, one  $^{15}\text{N}$  atom in the  $^{45}\text{N}_2\text{O}$  and two  $^{15}\text{N}$  atoms in the  $^{46}\text{N}_2\text{O}$ ), and  $R_{^{14}\text{N}-\text{N}_2\text{O}}$  ( $\text{pmol N l}^{-1} \text{d}^{-1}$ ) is the calculated production of  $^{14}\text{N}-\text{N}_2\text{O}$  based on the atom fractions of  $^{14}\text{N}$  ( $f_{14}$ ) and  $^{15}\text{N}$  ( $f_{15}$ ) of the substrate pool. The production of  $\text{N}_2\text{O}$  from one substrate is defined as the sum of  $R_{^{15}\text{N}-\text{N}_2\text{O}}$  and  $R_{^{14}\text{N}-\text{N}_2\text{O}}$ . The same equations were used to calculate the rates of  $\text{N}_2\text{O}$  production ( $R_{\text{NH}_4^+}$ ,  $R_{\text{NO}_2^-}$ ,  $R_{\text{NO}_3^-}$ ) from each of the individual tracers ( $^{15}\text{NH}_4^+$ ,  $^{15}\text{NO}_2^-$ ,  $^{15}\text{NO}_3^-$ ).

The gross  $\text{N}_2\text{O}$  production rate was derived from the sum of  $\text{NH}_4^+$  sourced,  $\text{NO}_2^-$  sourced and  $\text{NO}_3^-$  sourced  $\text{N}_2\text{O}$ . Therefore, the rate of gross  $\text{N}_2\text{O}$  production was calculated using equation (10):

$$R_{\text{gross}} = R_{\text{NH}_4^+} + R_{\text{NO}_2^-} + R_{\text{NO}_3^-} \quad (10)$$

where  $R_{\text{gross}}$  is the total  $\text{N}_2\text{O}$  production rate during our incubation ( $\text{pmol N l}^{-1} \text{d}^{-1}$ ). The errors of the  $\text{NH}_4^+$  sourced,  $\text{NO}_2^-$  sourced and  $\text{NO}_3^-$  sourced  $\text{N}_2\text{O}$  rate are based on the increase of  $\text{N}_2\text{O}$  of our incubation in the 2015 cruise (duplicates), 2018 cruise (triplicates) and 2019 cruise (triplicates), and propagation of the errors during the calculation using the equations listed above.

### Detection limits of rate measurements

For nitrification rate measurements, the detection limits depend on the concentration of the product pool and the fraction of  $^{15}\text{N}$  in the substrate pool during the incubation<sup>66,67</sup>. As mentioned, the precision of  $\delta^{15}\text{N}-\text{NO}_x^-$  was better than  $\pm 0.2\%$ , and we here use three times the standard deviation as a reliable enrichment of  $^{15}\text{N}$  in the product pool. Therefore, we calculated a detection limit of  $0.04\text{--}0.16 \text{ nmol N l}^{-1} \text{d}^{-1}$  for nitrification. Similarly, for  $\text{N}_2\text{O}$  production rate, the precision of  $\delta^{15}\text{N}-\text{N}_2\text{O}$  and  $\delta^{18}\text{O}-\text{N}_2\text{O}$  was better than  $\pm 0.3\%$  and  $\pm 0.4\%$ , respectively, and we here use three times the standard deviation as a reliable enrichment of  $^{45}\text{N}_2\text{O}$  and  $^{46}\text{N}_2\text{O}$  in the product pool. Therefore, we calculated a detection limit of  $0.1\text{--}0.3 \text{ pmol N l}^{-1} \text{d}^{-1}$  for  $^{45}\text{N}_2\text{O}$  production rate and  $0.2\text{--}0.6 \text{ pmol N l}^{-1} \text{d}^{-1}$  for  $^{46}\text{N}_2\text{O}$  production rate in  $^{15}\text{NH}_4^+$  tracer incubation;  $0.1\text{--}0.7 \text{ pmol N l}^{-1} \text{d}^{-1}$  for  $^{45}\text{N}_2\text{O}$  production rate and  $0.2\text{--}1.0 \text{ pmol N l}^{-1} \text{d}^{-1}$  for  $^{46}\text{N}_2\text{O}$  production rate in  $^{15}\text{NO}_2^-$  tracer incubation; and  $0.1\text{--}3.0 \text{ pmol N l}^{-1} \text{d}^{-1}$  for  $^{45}\text{N}_2\text{O}$  production rate and  $0.2\text{--}5.0 \text{ pmol N l}^{-1} \text{d}^{-1}$  for  $^{46}\text{N}_2\text{O}$  production rate in  $^{15}\text{NO}_3^-$  tracer incubation.

### Data availability

All data needed to evaluate the conclusions in the paper are deposited in the Zenodo database and can be accessed through <https://doi.org/10.5281/zenodo.6867932>.

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### Author contributions

X.S.W., M.D. and S.-J.K. conceived the study and designed the experiment. X.S.W., H.-X.S., K.L.C., W.Z., L.L., H.S. and K.Z. performed the experiment and measured the samples. X.S.W., H.-X.S., M.D., K.L.C., M.J.C., B.B.W. and S.-J.K. analysed the results and structured the manuscript. All authors contributed to the discussion of the results and editing of the manuscript.

### Competing interests

The authors declare no competing interests.

### Additional information

**Extended data** is available for this paper at <https://doi.org/10.1038/s41561-022-01090-2>.

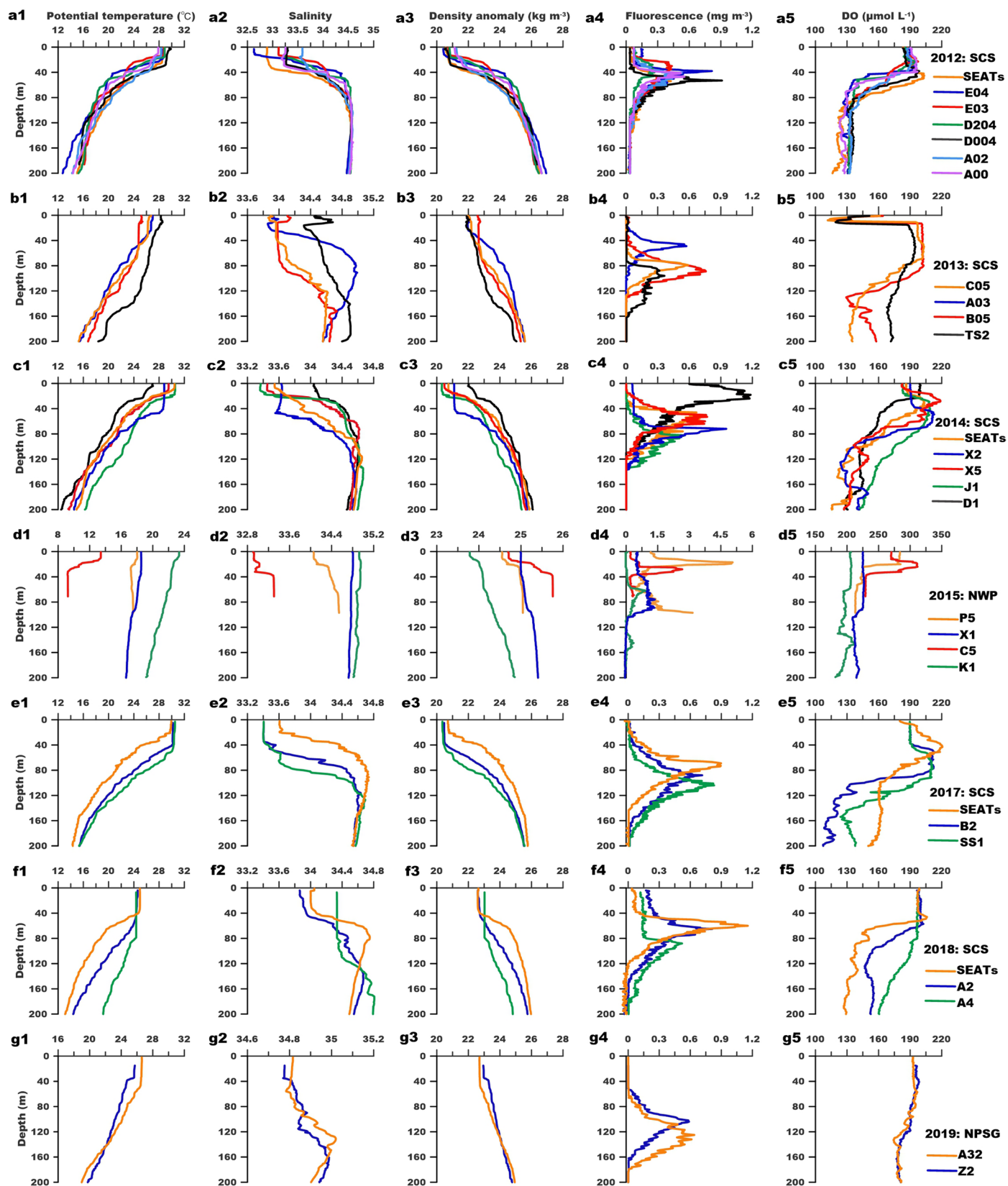
**Supplementary information** The online version contains supplementary material available at <https://doi.org/10.1038/s41561-022-01090-2>.

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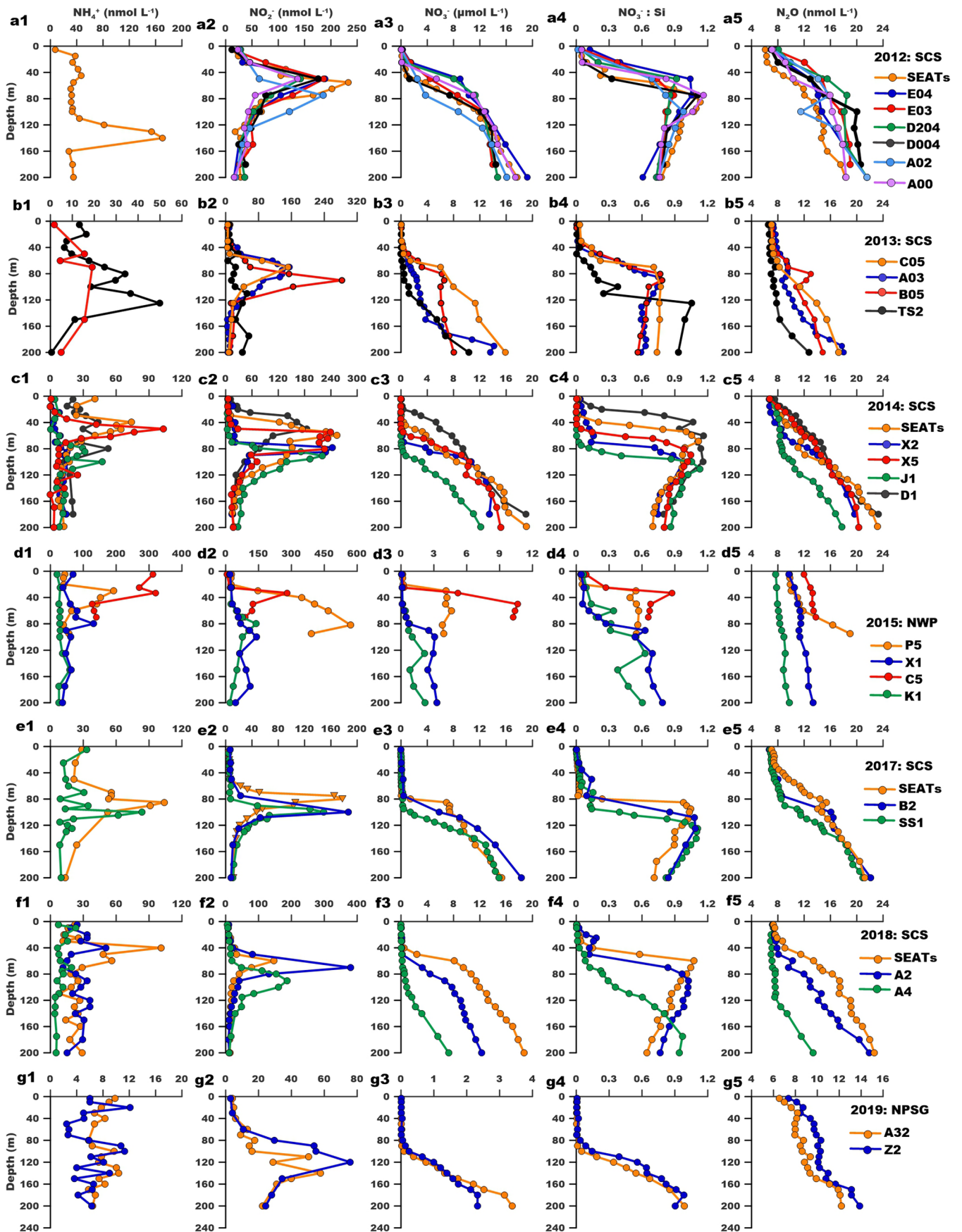
**Peer review information** *Nature Geoscience* thanks Christopher Somes and the other, anonymous, reviewer(s) for their contribution to the peer review of this work. Primary Handling Editor: Tom Richardson, in collaboration with the *Nature Geoscience* team.

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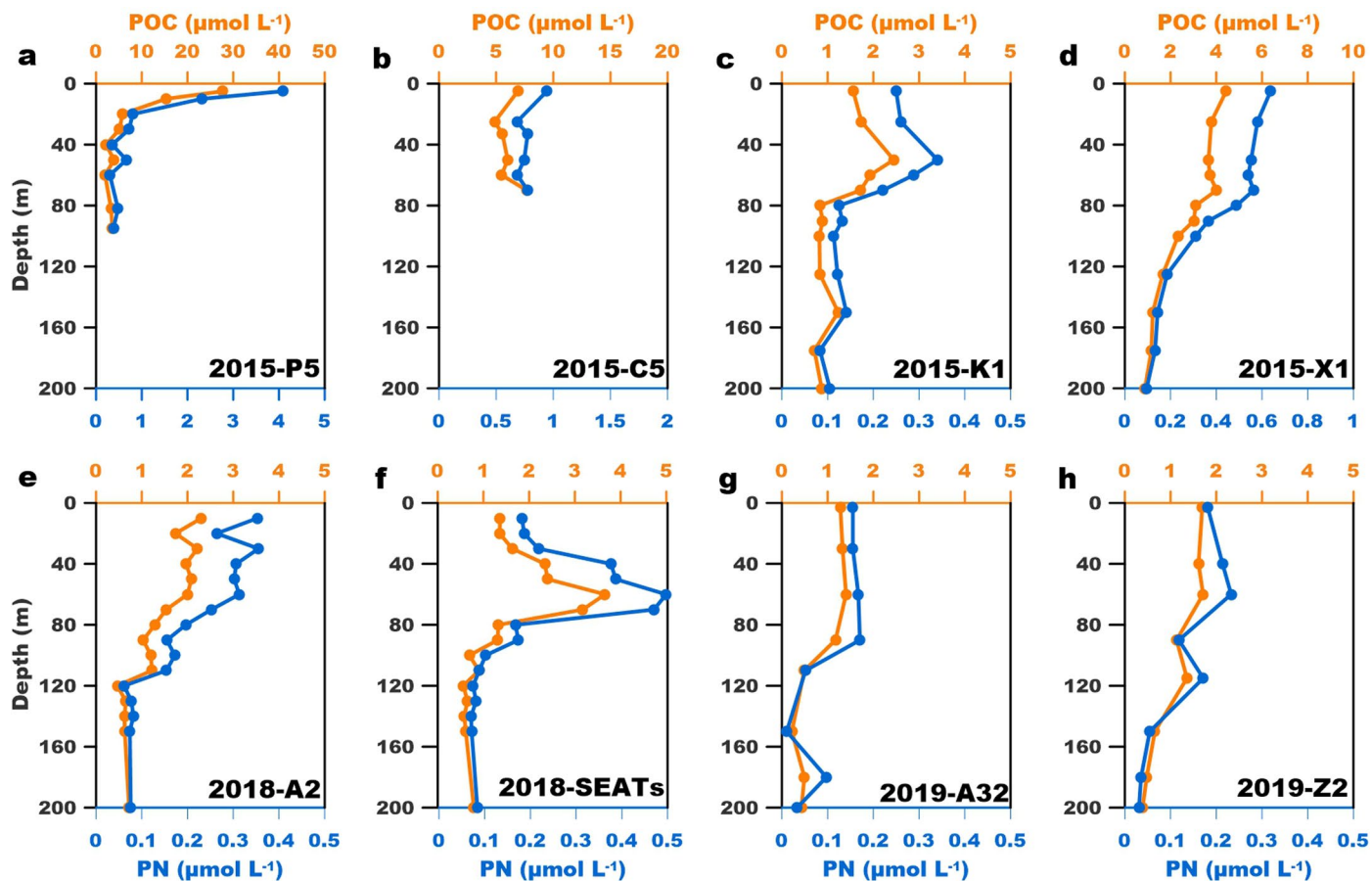




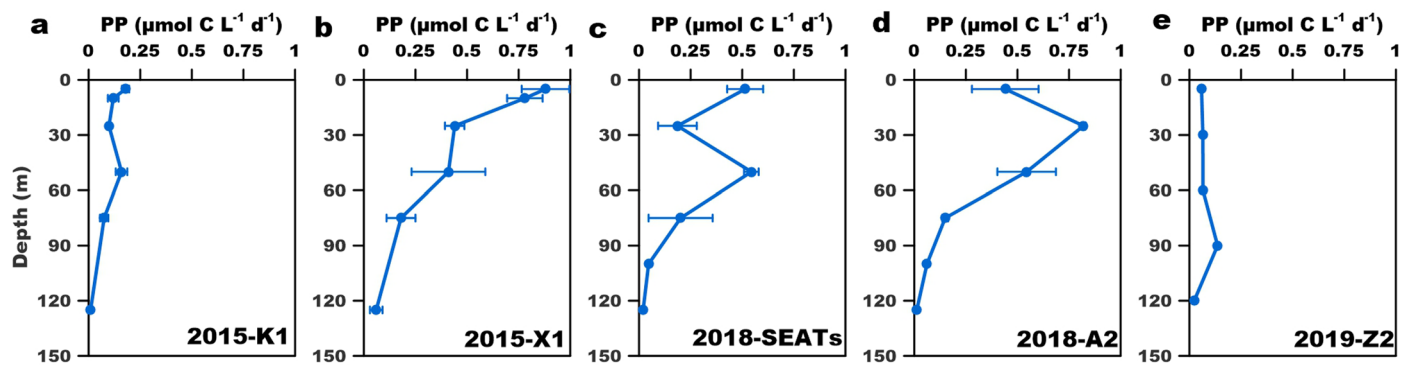
**Extended Data Fig. 1 | Hydrological and biological characteristics at sampling stations.** a1–g1, Potential temperature. a2–g2, Salinity. a3–g3, Potential density anomaly. a4–g4, Fluorescence. a5–g5, DO at the South China Sea (SCS); Northwest Pacific (NWP) and North Pacific Subtropical Gyre (NPSG). Each row of plots depicts hydrological and biological parameters from a single cruise.



**Extended Data Fig. 2 | Depth profiles of concentrations of nutrients and N<sub>2</sub>O at sampling stations.** a1-g1, NH<sub>4</sub><sup>+</sup>. a2-g2, NO<sub>2</sub><sup>-</sup>. a3-g3, NO<sub>3</sub><sup>-</sup>. a4-g4, NO<sub>3</sub><sup>-</sup>/Si. a5-g5, N<sub>2</sub>O at the South China Sea (SCS); Northwest Pacific (NWP) and North Pacific Subtropical Gyre (NPSG).. Each row depicts nutrient and N<sub>2</sub>O concentrations from a single cruise.



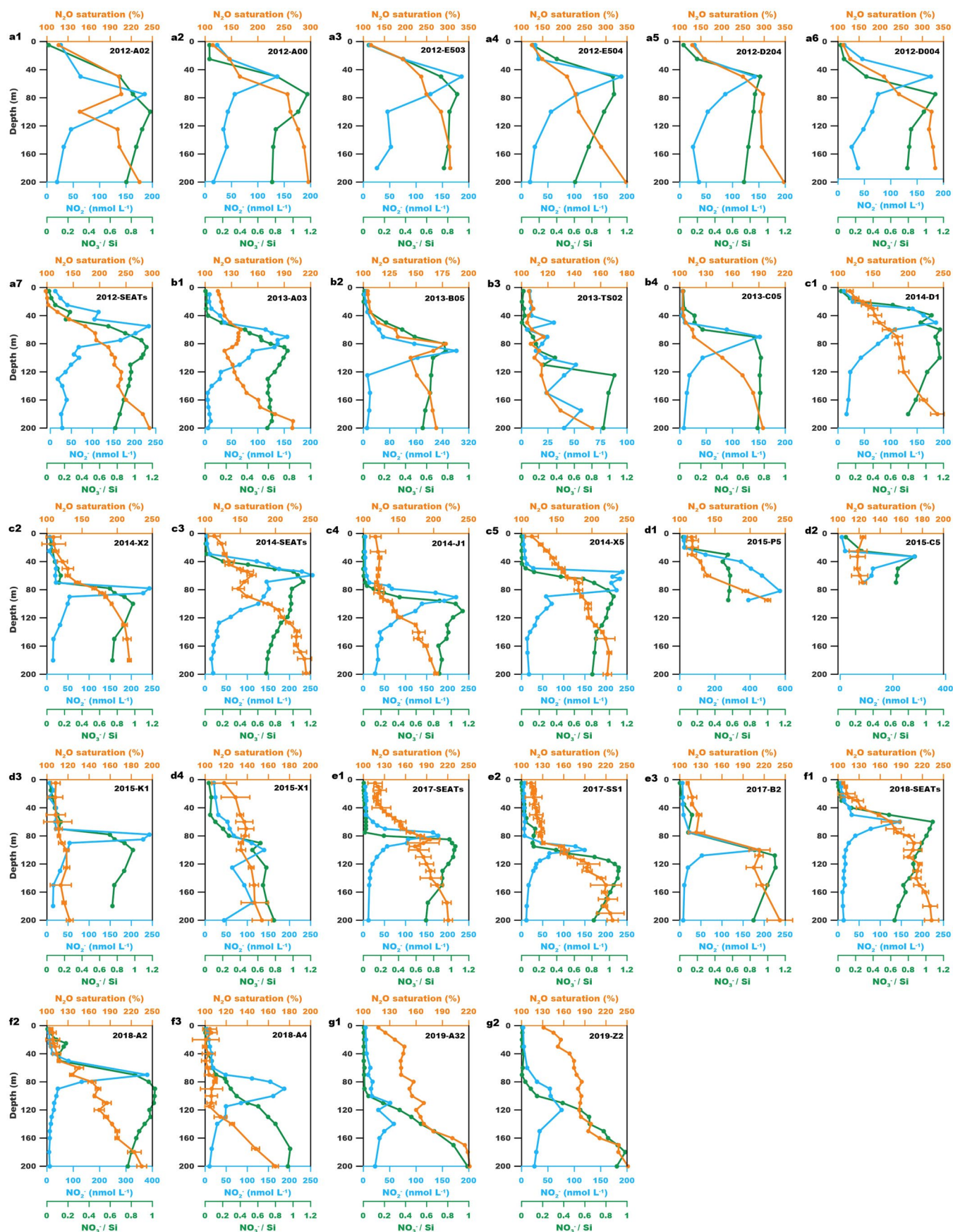
**Extended Data Fig. 3 | Depth profiles of POC and PN concentrations.** a-d, Station P5, C5, K1 and X1 in the subtropical Northwest Pacific in 2015. e-f, Station A2 and SEATs in the South China Sea in 2018. g-h, A32 and Z2 in the North Pacific Subtropical Gyre in 2019. POC concentrations are depicted with orange circles and PN concentrations are blue circles.



**Extended Data Fig. 4 | Depth profiles of primary production (PP) rate.** a-b, Station K1 and X1 in the subtropical Northwest Pacific in 2015. c-d, Station SEATs and A2 in the South China Sea in 2018. e, Station Z2 in the North Pacific

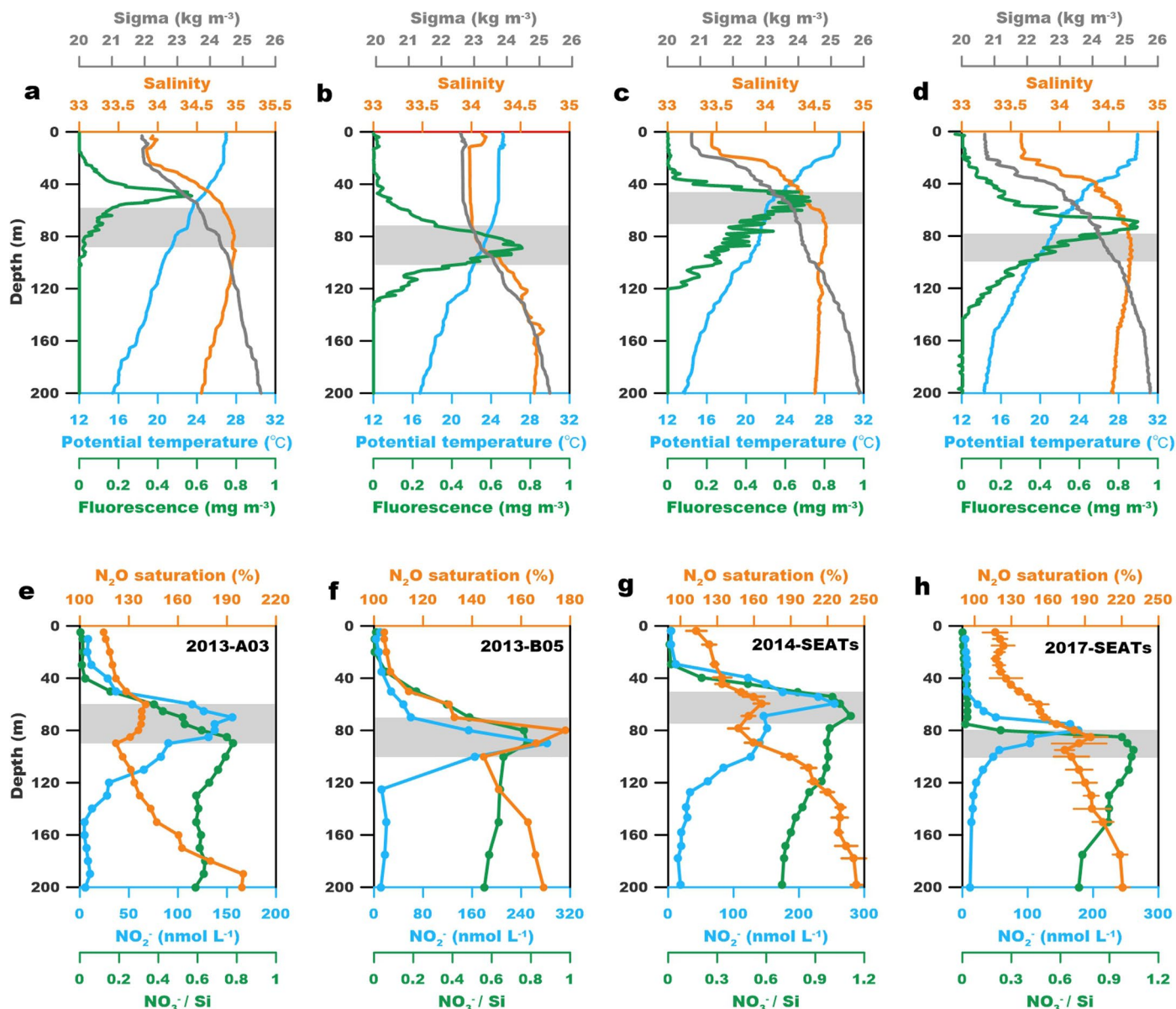
Subtropical Gyre in 2019. Data are presented as mean rates  $\pm$  standard deviation of triplicate sample incubation in the SCS and NPSG, and duplicates in the 2015 NWP cruise, are smaller than the symbols where not visible.





**Extended Data Fig. 5 | Depth profiles of N<sub>2</sub>O saturation, NO<sub>2</sub><sup>-</sup> and NO<sub>3</sub><sup>-</sup>/Si at stations with distinctive shallow N<sub>2</sub>O accumulation in the epipelagic zone.** N<sub>2</sub>O saturation (orange dots); NO<sub>2</sub><sup>-</sup> concentration (blue dots); NO<sub>3</sub><sup>-</sup>/Si (green dots). a1–a7, 2012 cruise. b1–b4, 2013 cruise. c1–c5, 2014 cruise. d1–d4, 2015

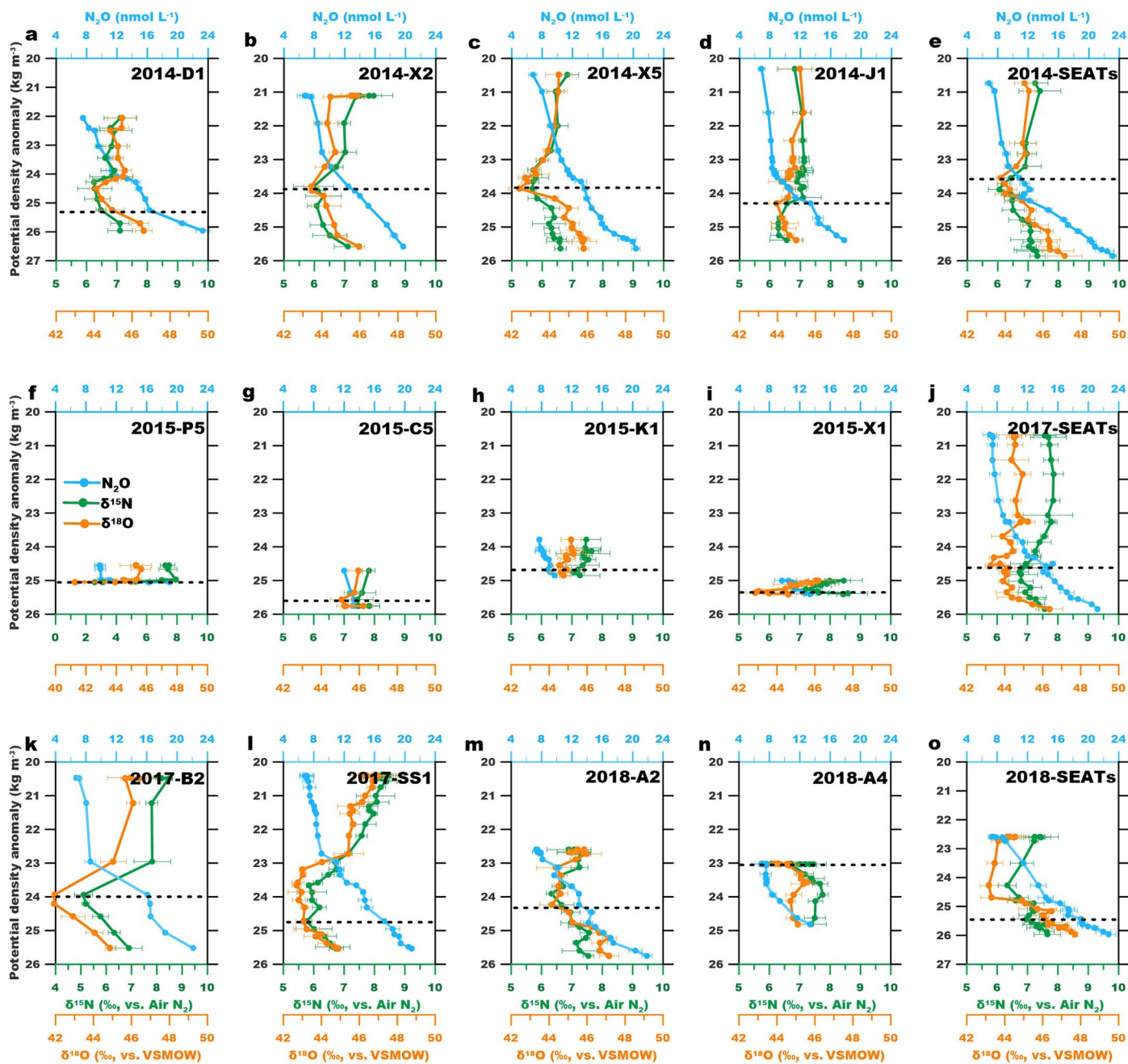
cruise. e1–e3, 2017 cruise. f1–f3, 2018 cruise. g1–g2, 2019 cruise. Data are presented as mean values ± standard deviation of triplicate sample measurement, and are smaller than the symbols where not visible.



**Extended Data Fig. 6 | Depth profiles of physical parameters and  $\text{N}_2\text{O}$  saturation at four stations with distinctive shallow  $\text{N}_2\text{O}$  accumulation in the SCS. a-d, The potential temperature (blue lines); salinity (orange lines); fluorescence (green lines) and potential density anomaly (gray lines). e-h,  $\text{N}_2\text{O}$  saturation (orange dots);  $\text{NO}_2^-$  concentration (blue dots);  $\text{NO}_3^-/\text{Si}$  (green dots)**

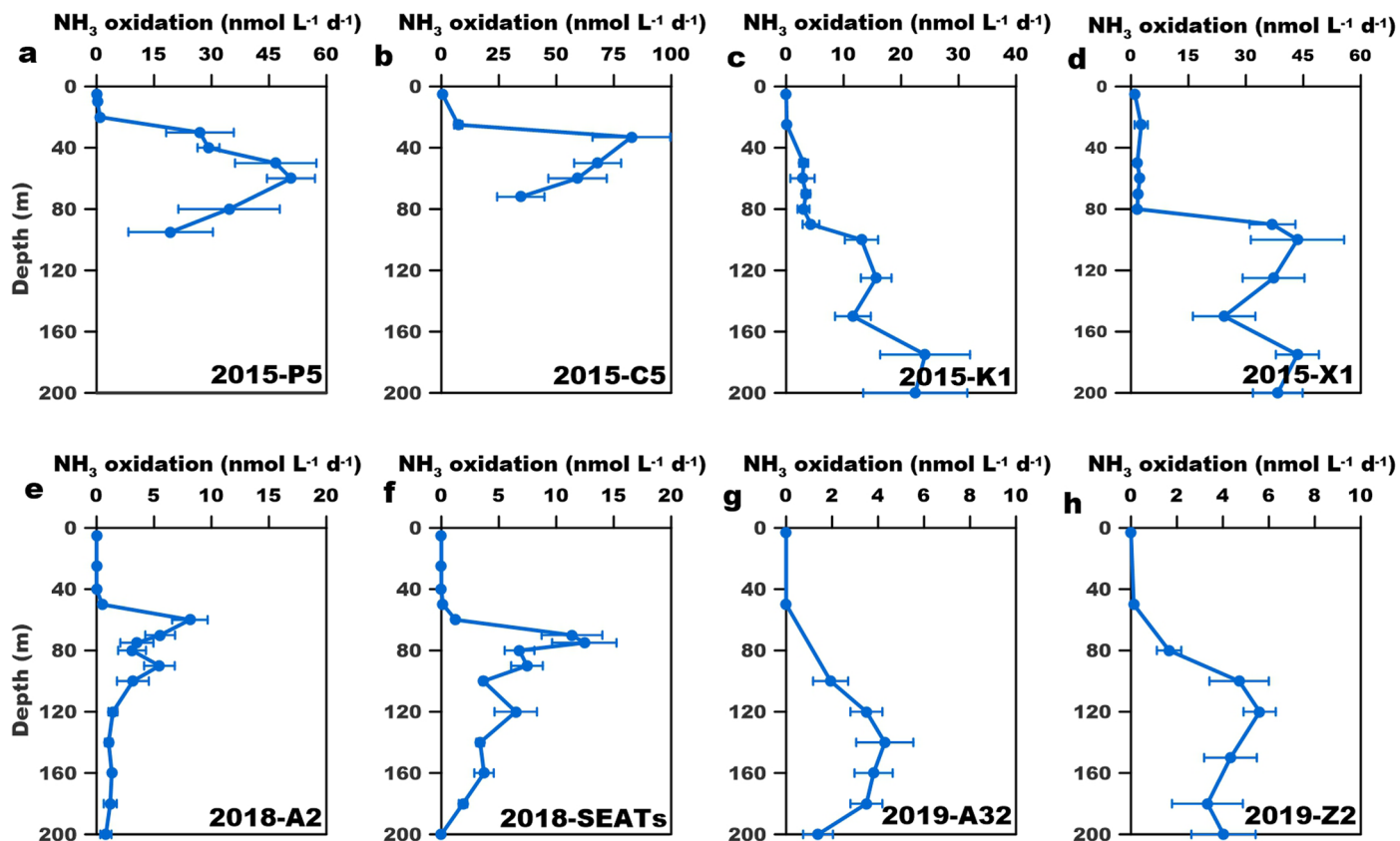
a, e, Station A03; b, f, Station B05; c, g, Station SEATs (2014); d, h, Station SEATs (2017). Gray bars mark the depth range with the distinctive  $\text{N}_2\text{O}$  accumulation. Data are presented as mean values  $\pm$  standard deviation of triplicate sample measurement, and are smaller than the symbols where not visible in panels g and h.





**Extended Data Fig. 7 | Concentration and dual isotopes of N<sub>2</sub>O plotted against potential density anomaly.** a–e, Station D1, X2, X5, J1, SEATs in the South China Sea (SCS) in 2014. f–i, Station P5, C5, K1, X1 in the Northwest Pacific in 2015. j–l, Station SEATs, B2, SS1 in the SCS in 2017. m–o, Station A2, A4, SEATs in the

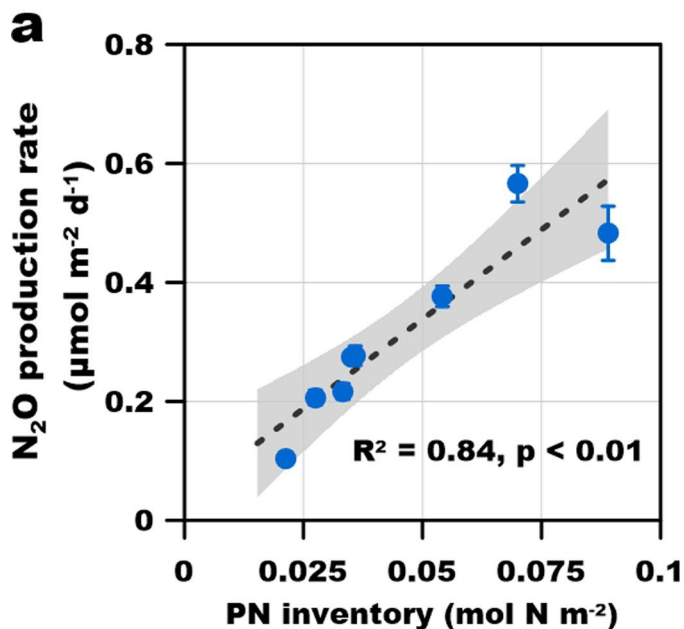
SCS in 2018. The dashed lines mark the depth with  $\delta^{15}\text{N-N}_2\text{O}$  isotope minima. Data are presented as mean values  $\pm$  standard deviation of triplicate sample measurement, and are smaller than the symbols where not visible.



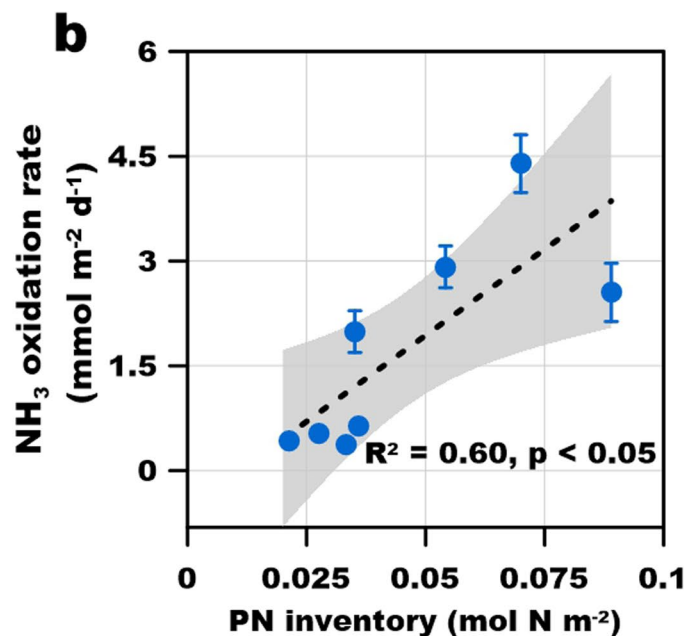
**Extended Data Fig. 8 | Depth profiles of ammonia oxidation rates.** a-d, Station P5, C5, K1 and X1 in the Northwest Pacific (NWP) in 2015. e-f, Station SEATs and A2 in the South China Sea (SCS) in 2018. g-h, Station A32 and Z2 in the North Pacific

Subtropical Gyre (NPSG). Data are presented as mean rates  $\pm$  standard deviation of triplicate sample incubation in the SCS and NPSG, and duplicates in the 2015 NWP cruise, are smaller than the symbols where not visible.

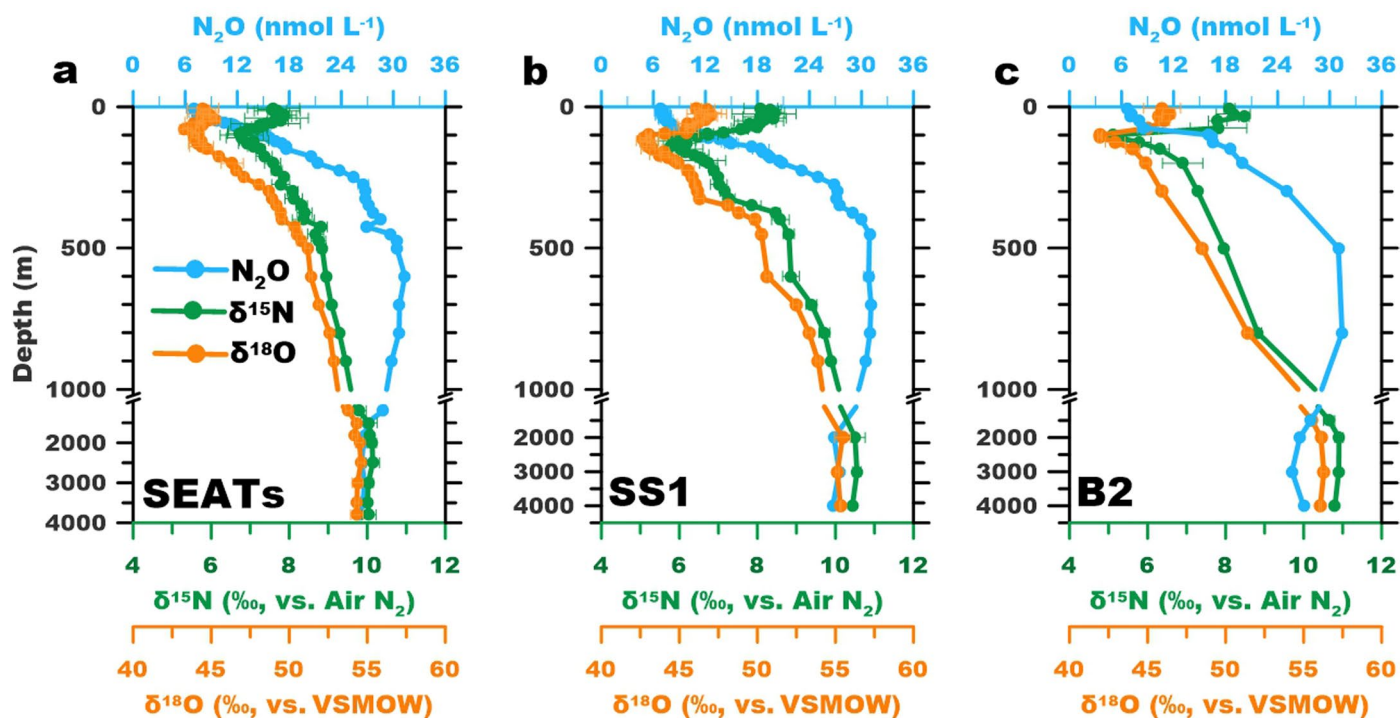




**Extended Data Fig. 9 | Relationships between the depth-integrated N<sub>2</sub>O production rates, ammonia oxidation rates and PN inventories.** a, Depth-integrated (0–200 m for the stations in open ocean and surface to bottom for the shelf stations) N<sub>2</sub>O production rate vs. particulate nitrogen (PN) inventory. b, Depth-integrated ammonia oxidation rates vs. PN inventories. The blue dots denote the measured rates and PN inventory at station P5, C5, K1, X1, A2,



SEATs (2018), A32 and Z2. Data are presented as the depth-integrated rates  $\pm$  propagated standard deviation of duplicates in the 2015 Northwest Pacific cruise and triplicates in the South China Sea and North Pacific Subtropical Gyre, and are smaller than the symbols where not visible. The black lines and gray shadows show linear regressions and the 95% confidence intervals, respectively.



**Extended Data Fig. 10 | Depth profiles of concentrations and dual isotopes of  $N_2O$  for sampling sites.** a-c, Station SEATs, SS1 and B2 in the South China Sea in 2017. Data are presented as mean values  $\pm$  standard deviation of triplicate sample measurement, and are smaller than the symbols where not visible.